

ME 40 Midterm 2 Solutions

① $T_1 = 20^\circ\text{C}$
 $P_1 = 100 \text{ kPa}$
 $T_4 = 1400^\circ\text{C}$
 $T_5 = 700^\circ\text{C}$
 $P_5 = 100 \text{ kPa}$

Working fluid = Helium

$\Delta P_{23} = -120 \text{ kPa}$
 $\dot{V}_1 = 4.0 \text{ m}^3/\text{s}$

$r_p = 14.1$
 $\eta_c = 0.91$

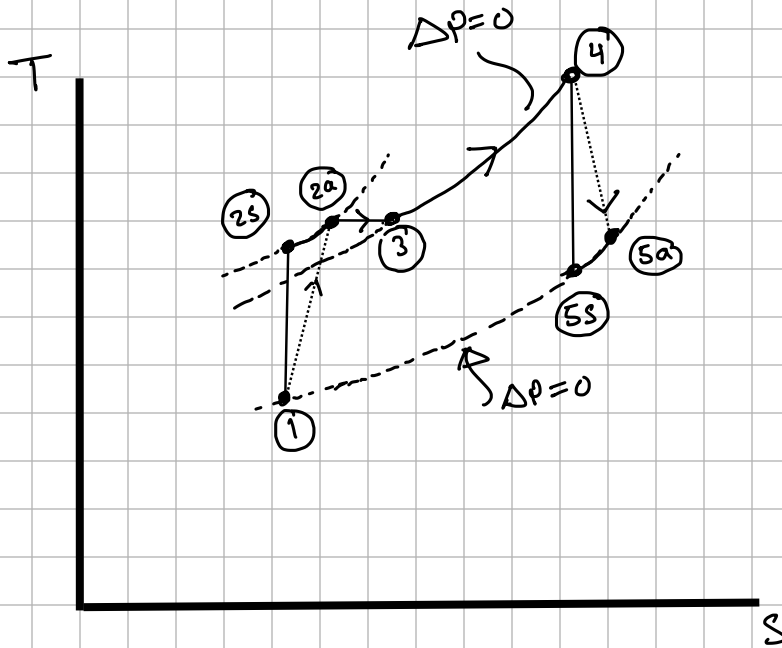


Table A-2: Helium

$R = 2.0769 \text{ kJ/kg}\cdot\text{K}$
 $C_p = 5.1926 \text{ kJ/kg}\cdot\text{K}$
 $k = 1.667$

Find mass flow rate: $\dot{m} = \frac{\dot{V}_1}{v_1}$

$\dot{m} = \frac{4.0}{6.085} = 0.657$

Ideal gas Law: $v_1 = \frac{RT_1}{P_1}$

$v_1 = \frac{(2.0769)(20+273)}{100}$

$v_1 = 6.085 \text{ m}^3/\text{kg}$

b) Find Isentropic State 2: Use Ideal gas isentropic relations

$\left(\frac{T_{2s}}{T_1}\right) = \left(\frac{P_2}{P_1}\right)^{(k-1)/k} = (r_p)^{(k-1)/k} \Rightarrow \left(\frac{T_{2s}}{293}\right) = (14.1)^{(1.667-1)/1.667}$

$\hookrightarrow T_{2s} = 844.68 \text{ K}$

$P_2 = 14.1 P_1 = 1410 \text{ kPa}$

↳ Find Actual state 2

$$\eta_c = \frac{h_{2s} - h_1}{h_{2a} - h_1} = \frac{c_p(T_{2s} - T_1)}{c_p(T_{2a} - T_1)} = \frac{844.68 - 293}{T_{2a} - 293} = 0.91$$

$$T_{2a} = 899.24 \text{ K}$$

$$\dot{W}_c = \dot{m}(h_{2a} - h_1) = \dot{m}c_p(T_{2a} - T_1)$$

$$= (0.6573)(5.1926)(899.24 - 293)$$

$$\dot{W}_c = 2069.16 \text{ kW}$$

c) For a throttling valve: $h_3 = h_{2a}$, $P_3 = P_2 - 120 = 1290 \text{ kPa}$

$$\rightarrow \Delta h = c_p \Delta T \rightarrow \Delta T = 0 \rightarrow T_3 = T_{2a} = 899.24 \text{ K}$$

d) $\dot{Q}_{in} = \dot{m}(h_4 - h_3) = \dot{m}c_p(T_4 - T_3)$ $T_4 = 1400 + 273 = 1673 \text{ K}$

$$= (0.6573)(5.1926)(1673 - 899.24)$$

$$\dot{Q}_{in} = 2640.9 \text{ kW}$$

e) Find Isentropic state 5: $P_4 = P_3 = 1290 \text{ kPa}$

$$\left(\frac{T_{5s}}{T_4}\right) = \left(\frac{P_5}{P_4}\right)^{(k-1)/k} = \left(\frac{100}{1290}\right)^{(1.667-1)/1.667}$$

$$\rightarrow T_{5s} = 601.35 \text{ K}, T_{5a} = 973 \text{ K}$$

$$\eta_T = \frac{h_4 - h_{5a}}{h_4 - h_{5s}} = \frac{c_p(T_4 - T_{5a})}{c_p(T_4 - T_{5s})} = \frac{(1673 - 973)}{(1673 - 601.35)} = 0.65$$

$$\eta_T = 0.65$$

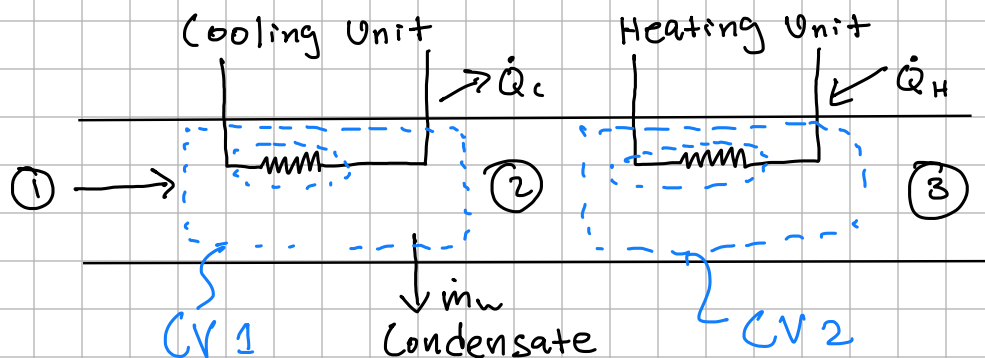
$$f) \dot{W}_{net} = \dot{W}_T - \dot{W}_c \quad \dot{W}_T = \dot{m}(h_4 - h_{5a}) = \dot{m}c_p(T_4 - T_{5a})$$
$$= 0.6573(5.1926)(1673 - 973)$$

$$\dot{W}_T = 2389.17 \text{ kW}$$

$$\dot{W}_{net} = 2389.17 - 2069.16 = 320.01 \text{ kW} = \dot{W}_{net}$$

$$\eta = \frac{\dot{W}_{net}}{\dot{Q}_{in}} = \frac{320.01}{2640.9} = 0.121 \rightarrow \eta = 12.1\%$$

②



$$T_{DB,1} = 34^\circ\text{C}$$

$$\phi_1 = 80\%$$

$$\dot{V}_1 = 5.0 \text{ m}^3/\text{min}$$

$$T_{WB,2} = 6.0^\circ\text{C}$$

$$T_{DB,3} = 21^\circ\text{C}$$

a) $\dot{m} = \frac{\dot{V}_1}{v_{da,1}}$ $v_{da,1} = 0.9075 \text{ m}^3/\text{kg}$ (approximation from psych chart)

$$\dot{m} = \frac{5.0/60}{0.9075} = 0.0918 \frac{\text{kg}}{\text{s}} = \dot{m}$$

b) To find \dot{Q}_c and \dot{Q}_h , set up two control volumes
 \hookrightarrow 1st CV around cooling coil

Mass Conservation: air - $\dot{m}_{da1} = \dot{m}_{da2} = \dot{m}_{da}$

Water - $\dot{m}_{da}w_1 = \dot{m}_{da}w_2 + \dot{m}_w$

SSSF: $\dot{m}_{da}h^*_1 = \dot{m}_{da}h^*_2 + \dot{m}_wh_w + \dot{Q}_c$

At Point 2: $\phi = 100\% \rightarrow T_{DB2} = T_{WB2} = 6^\circ\text{C}$

$\hookrightarrow h_w = h_f @ T = 6^\circ\text{C} = 25.22 \text{ kJ/kg}$ (Table A-4)

From Psych Chart:

$$h^*_1 = 104.5 \text{ kJ/kg}$$

$$h^*_2 = 20.5 \text{ kJ/kg}$$

$$w_1 = 27.3 \text{ g/kg air}$$

$$w_2 = 5.75 \text{ g/kg air}$$

$$\text{Water Cons: } (0.0918)(0.0273) = (0.0918)(0.00575) + \dot{m}_w$$

$$\hookrightarrow \dot{m}_w = 0.00197 \text{ kg/s}$$

$$\text{SSSF: } (0.0918)(104.5) = (20.5)(0.0918) + (0.00197)(25.22) + \dot{Q}_c$$

$$\hookrightarrow \dot{Q}_c = 7.66 \text{ kW}$$

→ Now set up 2nd Control Volume around the heating coil

Conservation of Mass: air: $\dot{m}_{da2} = \dot{m}_{da3} = \dot{m}_{da}$

Water: $\dot{m}_{da}w_2 = \dot{m}_{da}w_3 \rightarrow w_2 = w_3 = 5.75 \text{ g/kg}$

$$\text{SSSF: } \dot{m}_{da}h^*_2 + \dot{Q}_H = \dot{m}_{da}h^*_3$$

$$\hookrightarrow \dot{Q}_H = \dot{m}_{da}(h^*_3 - h^*_2)$$

From psych chart: $h^*_3 = 36.0 \text{ kJ/kg}$

$$\hookrightarrow \dot{Q}_H = (0.0918)(36.0 - 20.5) = 1.423 \text{ kW} = \dot{Q}_H$$

c) $T_{DB,2} = T_{WB,2} = 6.0^\circ\text{C}$

ASHRAE PSYCHROMETRIC CHART NO. 1

NORMAL TEMPERATURE SEA LEVEL

BAROMETRIC PRESSURE 101.325 kPa.



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