

Write your name here:

Solutions

Instructions:

- Answer all questions to the best of your abilities. Be sure to write legibly and state your answers clearly.
- The point values for each question are indicated.
- You are not allowed to use notes, friends, phones, etc.
- You can use calculators.
- There are a total of 100 points.
- Feel free to ask me questions, but only for clarification purposes.

This exam can contribute up to 20% of the points accrued toward your final grade, as stated in the syllabus. Good luck. I sincerely hope you all do really well.

-Prof. Chrzan

1. An electron bound to a hydrogen atom is described by the wavefunction:

$$\Psi(\mathbf{r}, t) = \frac{i}{4} \exp\left(\frac{-iE_1 t}{\hbar}\right) \psi_{100}(\mathbf{r}) - \frac{\sqrt{15}}{4} \exp\left(\frac{-iE_4 t}{\hbar}\right) \psi_{431}(\mathbf{r}),$$

with $\psi_{nlm}(\mathbf{r})$ an eigenfunction of the Hamiltonian, \hat{H} , squared angular momentum, \hat{L}^2 , and z-component of the angular momentum, \hat{L}_z .

(a) [10 points] Give the expectation value of the z-component of the angular momentum, \hat{L}_z , for the electron?

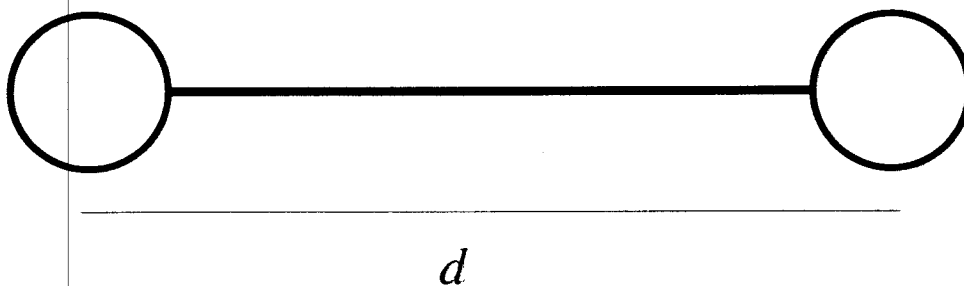
$$\langle \hat{L}_z \rangle = \frac{15}{16} \hbar \quad \left(= \sum_{n,l,m} |b_{nlm}|^2 \hbar m \right)$$

(b) [10 points] Give the expectation value of the total energy. (You can express this in terms of the energies E_1 and E_4 .)

$$\langle \hat{H} \rangle = \langle E \rangle = \frac{1}{16} E_1 + \frac{15}{16} E_4 \quad \left(= \sum_{nlm} |b_{nlm}|^2 E_n \right)$$

(c) [10 points] Give the expectation value of the angular momentum squared, \hat{L}^2 ?

$$\langle \hat{L}^2 \rangle = \frac{15}{16} \hbar^2 \cdot 3 \cdot 4 = \frac{45}{4} \hbar^2 \quad \left(= \sum_{nlm} |b_{nlm}|^2 \hbar^2 l(l+1) \right)$$



(2) The Hamiltonian matrix for an electron bound to a dimer with bond length d , determined using a basis composed of the S states of the bare atoms, is given by:

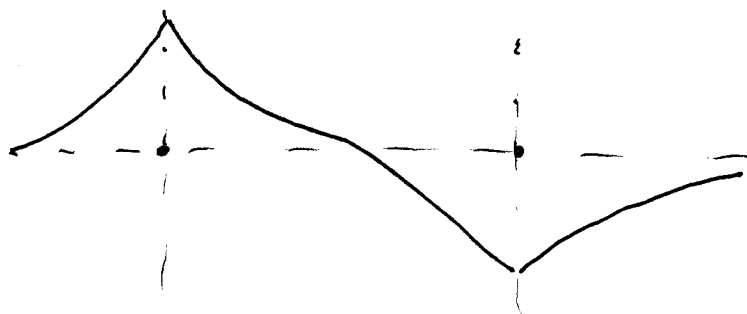
$$\mathbf{H} = \begin{pmatrix} E_0 & -V_0/d \\ -V_0/d & E_0 \end{pmatrix}$$

Here, E_0 is the energy level for an electron in a bare atom. All energies are measured in eV, and all distances are measured in Ångstroms.

(a) [20 points] What energies may one measure for an electron bound to the dimer? (Give your answer as a function of d .)

$$\begin{vmatrix} E_0 - E & -V_0/d \\ -V_0/d & E_0 - E \end{vmatrix} = 0 \quad \Rightarrow \quad E = E_0 \pm V_0/d$$

(b) [10 points] Plot the eigenfunction associated with the *antibonding* state. (Assume the atomic basis states are $\psi_{100}(\mathbf{r})$ states.) Plot this along the line connecting the dimer.



(c) [10 points] The driving force for bond formation is given by the reduction in electron energy upon formation of the dimer. This is counteracted by a repulsive energy. Suppose that this repulsive energy is given by a form similar to that for the Lennard-Jones potential, but with a smaller exponent (2 instead of 12), and no factor of 4:

$$V_{\text{repulsive}} = \varepsilon \left(\frac{\sigma}{d} \right)^2$$

Compute the zero temperature equilibrium bond length of the dimer in terms of V_0 , ε , and σ . Assume that there are two electrons bound to the dimer. Also assume that the electron-electron interactions are already included in the definition of V_0 . Using this bond length, compute the cohesive energy of the dimer. That is, compute the amount the energy is reduced upon bringing the atoms from $d = \infty$ to $d = \text{equilibrium bond length}$.

$$E(d) = \underbrace{-2 \frac{V_0}{d}}_{\substack{\uparrow \\ \text{from} \\ \text{electrons}}} + \underbrace{\varepsilon \left(\frac{\sigma}{d} \right)^2}_{\substack{\uparrow \\ \text{repulsive part}}}$$

$$\left. \frac{dE}{dd} \right|_{d=d_0} = 0 = \frac{2V_0}{d_0^2} - \frac{2\varepsilon}{d_0} \left(\frac{\sigma}{d_0} \right)^2$$

$$\Rightarrow \frac{V_0}{d_0^2} = \frac{\varepsilon}{d_0} \frac{\sigma^2}{d_0^2}$$

$$d_0 = \frac{\varepsilon \sigma^2}{V_0}$$

\uparrow equilibrium bond length

$$E(d_0) = \frac{-2V_0}{\left(\frac{\varepsilon \sigma^2}{V_0} \right)} + \frac{\varepsilon \sigma^2}{\left(\frac{\varepsilon \sigma^2}{V_0} \right)^2} = \frac{-2V_0^2}{\varepsilon \sigma^2} + \frac{V_0^2}{\varepsilon \sigma^2} = -\frac{V_0^2}{\varepsilon \sigma^2}$$

(3) Suppose that the interaction between two atoms is given by the Lennard-Jones potential:

$$V(d) = 4\epsilon \left[\left(\frac{\sigma}{d} \right)^{12} - \left(\frac{\sigma}{d} \right)^6 \right]$$

In this problem, you will consider an infinite one-dimensional lattice with one atom per site. You may need the sums:

$$\sum_{n=-\infty}^{\infty} \left(\frac{1}{n} \right)^{12} = \frac{1382 \pi^{12}}{638512875} = 2.00049 = A_{12}$$

$$\sum'_{n=-\infty} \left(\frac{1}{n} \right)^6 = \frac{2 \pi^6}{945} = 2.03469 = A_6$$

The primes on the sums indicate that they excluded the $n = 0$ term. Choose $\epsilon = 4.0$ eV, and $\sigma = 2.0$ Å.

(a) [15 points] Find the equilibrium lattice parameter for the chain of atoms.

$$V(a) = \frac{4\epsilon}{2} \sum_n \sum_{n'} \left[\left(\frac{\sigma}{a|n-n'|} \right)^{12} - \left(\frac{\sigma}{a|n-n'|} \right)^6 \right]$$

$$= 2\epsilon N \sum_{n'} \left[\left(\frac{\sigma}{a} \right)^{12} \left(\frac{1}{|n'|} \right)^{12} - \left(\frac{\sigma}{a} \right)^6 \left(\frac{1}{|n'|} \right)^6 \right]$$

$$= 2\epsilon N \left[A_{12} \left(\frac{\sigma}{a} \right)^{12} - A_6 \left(\frac{\sigma}{a} \right)^6 \right]$$

$$\left. \frac{dV}{da} \right|_{a=a_0} = 2\epsilon N \left[-12 \frac{A_{12}}{a_0} \left(\frac{\sigma}{a_0} \right)^{12} + \frac{6A_6}{a_0} \left(\frac{\sigma}{a_0} \right)^6 \right] = 0$$

$$\Rightarrow 2A_{12} \left(\frac{\sigma}{a_0} \right)^6 = A_6$$

$$a_0 = \left(\frac{2A_{12}}{A_6} \right)^{1/6} \sigma = 2.23859 \text{ Å}$$

(b) [15 points] Find the cohesive energy per atom of the chain of atoms.

$$\begin{aligned}\frac{V(a_0)}{N} &= 2\varepsilon \left(A_{12} \left(\frac{\sigma}{a_0} \right)^{12} - A_6 \left(\frac{\sigma}{a_0} \right)^6 \right) \\ &= 2\varepsilon \left(A_{12} \left(\frac{A_6}{2A_{12}} \right)^2 - A_6 \left(\frac{A_6}{2A_{12}} \right) \right) \\ &= 2\varepsilon \left(\frac{A_6^2}{A_{12}} \right) \left[\frac{1}{4} - \frac{1}{2} \right] \\ &= -\frac{\varepsilon}{2} \frac{A_6^2}{A_{12}} = \cancel{-2.06111 \text{ eV}} \\ &\quad - 4.13895 \text{ eV}\end{aligned}$$