

Physics H7B - Spring 2002 (Bale)

Midterm Exam #1

One 3x5 note card is allowed

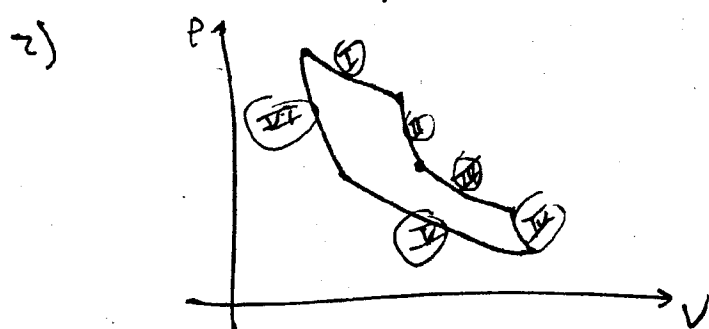
March 5, 2002

Problem 1 (25 points)

A system absorbs 200 J of energy from a reservoir at 400 K and also 300 J from a reservoir at 300 K. It interacts with a third reservoir whose temperature is T_3 . When the system returns to its original state, it has done 100 J of work.

- 1) What is the entropy change of the system for the complete cycle?
- 2) Assuming that the cycle is reversible, draw and label the P-V diagram.
- 3) What is T_3 ?

1) $\Delta S = 0$ since system returns to original state (2 pts)



- I isothermal expansion at 400K
 $\Delta S = \frac{200J}{400K} = \frac{Q_I}{T_I}$
- II adiabatic expansion
- III isothermal expansion at 300K
 $\Delta S = \frac{300J}{300K} = \frac{Q_{III}}{T_{III}}$
- IV adiabatic expansion
- V isothermal compression
 $\Delta S = \frac{Q_3}{T_3}$
- VI adiabatic compression

3) so $\Delta S = \frac{Q_I}{T_I} + \frac{Q_{III}}{T_{III}} + \frac{Q_3}{T_3} = 0$

since $\Delta T = 0 \rightarrow \Delta U = 0 = Q + W = Q_I + Q_{III} + Q_3 - W = 0$

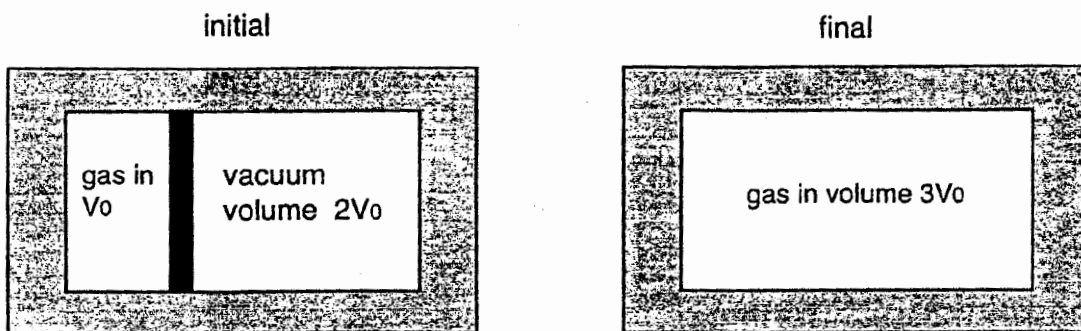
so $Q_3 = W - Q_I - Q_{III} = 100J - 200J - 300J$

$Q_3 = -400J$

so $\frac{200J}{400K} + \frac{300J}{300K} - \frac{400J}{T_3} = 0$

$\rightarrow T_3 = 267K$

Problem 2 (20 points)



A monatomic gas is contained in a volume V_0 , in an insulated container. A partition is removed and the gas expands to fill an additional volume $2V_0$. Calculate the change of entropy.

No heat flows through walls, so $dT = dU = 0$
 $T = \text{const}$

$$\Delta S = \frac{dQ}{T} \quad \text{but} \quad dQ = dW = p dV$$

$$\Delta S = \frac{dQ}{T} = p \frac{dV}{T}$$

ideal gas

$$pV = nk_B T$$

$$p = \frac{nk_B T}{V} \quad \text{where } n \equiv \text{number density}$$

$$\Delta S = nk_B T \frac{dV}{V T}$$

$$S_{\text{final}} - S_{\text{initial}} = nk_B \int_{V_0}^{3V_0} \frac{dV}{V}$$

$$= nk_B \ln V \Big|_{V_0}^{3V_0} = nk_B \ln \frac{3V_0}{V_0} = nk_B \ln 3$$

$$\text{so } \boxed{S_f - S_i = nk_B \ln 3}$$

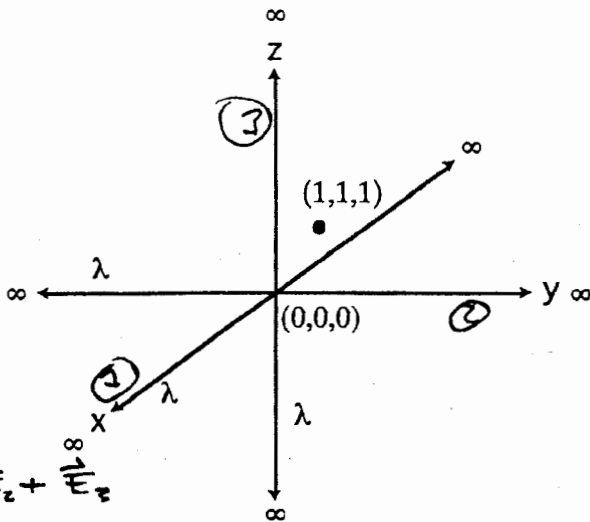
$$\text{or } NR \ln 3$$

$N = \# \text{ of molecules}$
 $R \equiv \text{gas constant}$

note entropy increases - free expansion of gas is irreversible!

Problem 3 (25 points)

Three infinite line charges, each with linear charge density λ , meet at the origin $(0,0,0)$ and are orthogonal to one another. Calculate the electric field magnitude and direction at the point $(1,1,1)$.



Use Gauss' Law ¹⁰ and superposition (pts)

for each line charge

$$\int E \cdot dA = 4\pi Q_{en}$$

$$E \cdot 2\pi r L = 4\pi \lambda L \rightarrow E = \frac{2\lambda}{r}$$

(5 pts)

$$\vec{E}_{total} = \vec{E}_1 + \vec{E}_2 + \vec{E}_3$$

① charge along \hat{x} : $\hat{r} = \frac{y}{r}\hat{y} + \frac{z}{r}\hat{z}$ so $\vec{E}_1 = \frac{2\lambda}{y^2+z^2} (y\hat{y} + z\hat{z})$

$$\vec{E}_1(1,1,1) = \lambda(\hat{y} + \hat{z})$$

② charge along \hat{y} : $\hat{r} = \frac{x}{r}\hat{x} + \frac{z}{r}\hat{z}$

$$\vec{E}_2 = \frac{2\lambda}{(x^2+z^2)} (x\hat{x} + z\hat{z})$$

$$\vec{E}_2(1,1,1) = \lambda(\hat{x} + \hat{z})$$

③ charge along \hat{z} : $\hat{r} = \frac{x}{r}\hat{x} + \frac{y}{r}\hat{y}$

$$\vec{E}_3 = \frac{2\lambda}{(x^2+y^2)} (x\hat{x} + y\hat{y})$$

$$\vec{E}_3(1,1,1) = \lambda(\hat{x} + \hat{y})$$

so $\vec{E}_{total} = \vec{E}_1 + \vec{E}_2 + \vec{E}_3$ at $(1,1,1)$

$$= \lambda(\hat{y} + \hat{z}) + \lambda(\hat{x} + \hat{z}) + \lambda(\hat{x} + \hat{y})$$

$$\vec{E} = 2\lambda(\hat{x} + \hat{y} + \hat{z})$$

$$|\vec{E}| = \sqrt{(2\lambda)^2 + (2\lambda)^2 + (2\lambda)^2} = \sqrt{12}\lambda = 2\sqrt{3}\lambda$$

$$\hat{E} = \frac{1}{\sqrt{3}}(\hat{x} + \hat{y} + \hat{z})$$

Problem 4 (30 points)

A coaxial cable carries a volume charge ρ on its inner cylinder (radius α) and a surface charge σ on the outer cylinder (radius β). The charges balance so that the total charge is zero. Calculate the electric field in the regions 1) $r < \alpha$, 2) $\alpha < r < \beta$, and 3) $r > \beta$. Plot E against r.

Use Gauss' Law with cylindrical symmetry (10 pts)

for $r < \alpha$: Gaussian surface within α

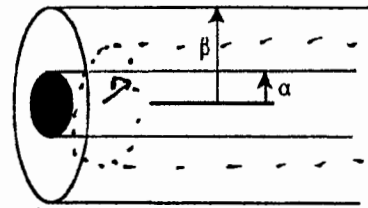


$$\int E \cdot dA = 4\pi Q_{enc}$$

$$E \cdot 2\pi r L = 4\pi (\rho \pi r^2 L)$$

$$\rightarrow \boxed{E = 2\pi \rho r} \quad (5 \text{ pts})$$

for $\alpha < r < \beta$: Gaussian surface between α & β



$$\int E \cdot dA = 4\pi Q_{enc} \quad \leftarrow \text{now entire inner cylinder is enclosed}$$

$$E \cdot 2\pi r L = 4\pi (\rho \pi \alpha^2 L)$$

$$\rightarrow \boxed{E = 2\pi \rho \frac{\alpha^2}{r}} \quad (5 \text{ pts})$$

for $r > \beta$: Gaussian surface outside coax

since total charge = 0

$$\int E \cdot dA = 0 \quad \rightarrow \boxed{E = 0} \quad (5 \text{ pts})$$

