

Sample MT # 2

$$\left. \begin{array}{l} \text{Total spin of lke: } S = 1 \\ \text{Total ang. mom. of lke: } l = 0 \end{array} \right\} j = 1 \text{ for lke}$$

j is conserved

electron has spin $\frac{1}{2}$: Thus the $S = 0, 1$

But $S = 0$ is ~~is~~ ruled out:

If $S = 0$ (antisymmetric), we would require a symmetric orbital state, i.e. even values of l . This would require an even value of j , which would not conserve j .

Thus $S = 1$ (symmetric)

This requires an antisymmetric orbital state, i.e. $l = 1, 3, \dots$

But since $j = 1, s = 1$, the only allowed value is $j = 1$

Thus $\boxed{l = 1, s = 1, j = 1}$

(a) For the symmetric case, we can have wavefunctions of the form

$$\psi_i(1)\psi_i(2)$$

or, for $j \neq i$,

$$\frac{1}{\sqrt{2}} (\psi_i(1)\psi_j(2) + \psi_j(1)\psi_i(2))$$

For the first case, there are n possible states ($n = 1, 2, \dots, n$). For the second case, i can take on n different values, and since $j \neq i$, j can take on $n-1$ values, for a total of $n(n-1)$ states. This double counts, however, since $(i = a, j = b)$ is equivalent to $(i = b, j = a)$. So the contribution due to $i \neq j$ terms is $n(n-1)/2$ [this is the same as n choose 2]. The total number of symmetric states is then,

$$n + \frac{1}{2}n(n-1) = \boxed{\frac{1}{2}n(n+1)}$$

(b) The $i = j$ case is no longer allowed, since it is necessarily symmetric. For $i \neq j$, we can antisymmetrize:

$$\frac{1}{\sqrt{2}} (\psi_i(1)\psi_j(2) - \psi_j(1)\psi_i(2))$$

So the number of states is just the number of $i \neq j$ states from part (a):

$$\boxed{\frac{1}{2}n(n-1)}$$

(c) The multiplicity of states for a particle with spin s is $n = 2s + 1$, so our expressions from above give that the ratio is

$$\frac{n(n+1)}{n(n-1)} = \frac{n+1}{n-1} = \frac{2s+2}{2s} = \boxed{\frac{s+1}{s}}$$

(a) Because the two electrons are in different orbital angular momentum states, they are inequivalent, so any of the two states accessible to the $2s$ electron may be paired with any of the six states available to the $2p$ electron, giving a degeneracy of $2 \times 6 = 12$. We are combining $l = 0$ with $l = 1$, which gives only $L = 1$. We are combining two spin-1/2 particles, so we can have $S = 0$ or $S = 1$. Again, the inequivalency removes any symmetrization requirements, so either combination is fine. The degeneracy of a given state is $2J + 1$. This gives

S	L	J	$^{2S+1}L_J$	Degeneracy
0	1	1	1P_1	3
1	1	2, 1, 0	$^3P_{2,1,0}$	$5 + 3 + 1 = 9$

So the total degeneracy is $3 + 9 = 12$, which agrees with our earlier calculation.

(b) Because the two electrons have different values of n , they are inequivalent. They are each in an $l = 1, s = 1/2$ state, which has degeneracy of 6, so the total degeneracy is $6 \times 6 = 36$. We are now combining two $l = 1$ states which gives us possible values for L of 2, 1, and 0. Again, the inequivalency of the electrons means that each of these may be paired with either spin state:

S	L	J	$^{2S+1}L_J$	Degeneracy
0	0	0	1S_0	1
0	1	1	1P_1	3
0	2	2	1D_2	5
1	0	1	3S_1	3
1	1	2, 1, 0	$^3P_{2,1,0}$	$5 + 3 + 1 = 9$
1	2	3, 2, 1	$^3D_{3,2,1}$	$7 + 5 + 3 = 15$

Summing these degeneracies, we get 36 as expected.

(c) There are only ten available states for an $l = 2, s = 1/2$ electron, and since we have ten electrons, there is only one possible configuration. Each spin up must be paired with a spin down, giving a total spin of $S = 0$. Similarly, we have total orbital angular momentum of $L = 0$. This gives us a term 1S_0 , which has degeneracy 1.

(d) We now have 9 d-electrons, which is equivalent to a single hole. There are ten ways that we can put a spin-1/2 hole into an $l = 2$ state, so we expect degeneracy 10. Since there is only one hole, $S = s = 1/2$, and $L = l = 2$. Our possible terms are then $^2D_{\frac{5}{2}, \frac{3}{2}}$, giving a degeneracy of $6 + 4 = 10$.

3. In the lectures we derived the anomalous Zeeman effect by writing

$$\vec{\mu}_{\text{atom}} = -g \frac{\mu_B}{\hbar} \vec{J}$$

where g is the Lande g factor. We neglected hyperfine interactions. One could account for them by further writing

$$\vec{\mu}_{\text{atom}} = -\frac{\mu_B}{\hbar} \left(g \vec{J} + g_I \frac{\mu_N}{\mu_B} \vec{I} \right) = -g_F \frac{\mu_B}{\hbar} \vec{F}.$$

Derive an expression for g_F in terms of g and g_I . Use the expression to calculate the weak field splitting of the H ground state. Draw a figure showing the energies with magnetic field strength and label with relevant quantum numbers.