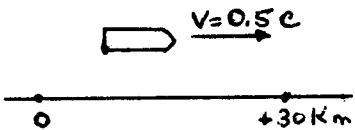


Physics 7C – Section 2  
 Fall Semester 2003  
 Second Midterm Exam (R. Marrus)  
 11/04/03

- 1) A single slit 1 mm wide is uniformly illuminated by light of wavelength  $\lambda = 589$  nm. A diffraction pattern is observed on a screen 3 m behind the slit. Throughout this calculation you may assume that the angles subtended at the slit are small.
- Find the distance between the first diffraction minima on each side of the central maximum.
  - The width of the slit is now doubled. Find: (I) The distance between the first diffraction minima on each side of the central maximum. (II) The factor by which the intensity of the central maximum changes.
  - The slit is now illuminated with a second source of light of wavelength  $\lambda'$ . The value of  $\lambda'$  is such that its second diffraction minimum on one side of the central maximum coincides with the third diffraction minimum produced by light of wavelength  $\lambda$ . Calculate  $\lambda'$ .
- 2) In an inertial frame fixed relative to the earth, a red light suddenly flashes at the origin and a white light suddenly flashes at a point + 30 km from the origin. In the earth's frame both flashes are simultaneous. A spaceship moves with a speed  $v = 0.5c$  parallel to the line between the 2 points at which the flashes occur. To an observer in the spaceship:
- What is the separation of the two lights?
  - What is the time interval between flashes?
  - Which flash occurs first?
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- 3) An x-ray photon of wavelength 0.0021 nm is Compton scattered by a free electron (initially at rest) through an angle of  $30^\circ$  from the incident direction. Find the kinetic energy in (eV) of the recoiling electron.
- 4) Determine the radius of the star Procyon B from the following data: the flux of starlight reaching us is  $1.7 \times 10^{-12}$  watt/m<sup>2</sup>, the distance of the star is 11 light years, and its' surface temperature is 6600°K. Assume the star radiates like a blackbody.
- Light of frequency  $10^{16}$  Hz is incident on a gas of hydrogen atoms in the ground state. What is the kinetic energy (in eV) of the emitted electrons?
  - A hydrogen atom makes a transition from  $n=3$  to  $n=2$  and emits a photon. Calculate the energy of the recoiling nucleus. You may assume the recoil energy is small compared to the photon energy.

$$h = 6.62 \times 10^{-34} \text{ joule-s} = 4.14 \times 10^{-15} \text{ eV-s}$$

$$c = 3 \times 10^8 \text{ m/sec.}$$

$$m_e (\text{electron mass}) = 9.1 \times 10^{-31} \text{ kg.} = 511 \text{ keV}/c^2$$

$$m_p (\text{proton mass}) = 1.67 \times 10^{-27} \text{ kg} = 938 \text{ MeV}/c^2$$

$$1 \text{ eV} = 1.6 \times 10^{-19} \text{ joule}$$

$$e (\text{electron charge}) = 1.6 \times 10^{-19} \text{ coulomb}$$

$$R (\text{Rydberg constant}) = 13.6 \text{ eV}$$

$$k = 1/(4\pi\epsilon_0) = 9 \times 10^9 \text{ MKS}$$

$$\sigma (\text{Stefan-Boltzmann constant}) = 5.67 \times 10^{-8} \text{ W/m}^2\text{-s}$$