

Chemistry 3A -
Midterm

chemistry 3A -
MIDTERM 2 - SPRING
2001 - FRECHET

Professor Jean Fréchet

Your full si

April 10, 2002

Print your

Middle)

Your SID_

Please check the section number and name of y

____ 161 Margot Paulick

____ 171 Danielle Dube

____ 181 Matt Pratt

____ 191 Laurie Schenkel

____ 111 Cathleen Yung

____ 121 Priya Sonik

____ 131 Paul Furuta

____ 141 Zach Fresco

____ 261 Aaron Stutz

____ 271 Vanessa Sun

____ 211 Jamey Kain

____ 221 Laura Anderson

____ 361 Reema Thalji

____ 371 Warren Wood

____ 471 Alex Kollias

____ 411 Andrew Chi

____ 421 Shahed Ghoghawala

____ 431 Mike Slater

____ 561 Jean Han

571 Ognjen Miljanic

____ 511 Joe Kwon

____ 521 Catherine Chan

____ 531 Olga Fedin

____ 541 Jason Serin

If you are making up an I-grade, indicate the semester you took 3A
____ and the Professor _____

Do not write in this box.

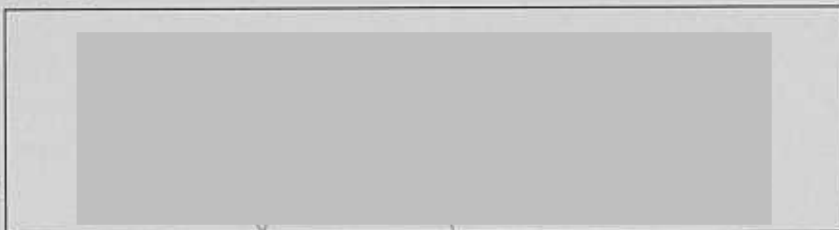
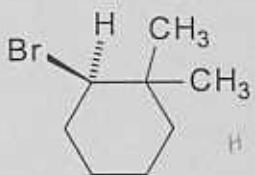
This exam has **10** pages; **make sure that you have them all.**

We will only grade answers that are in the designated spaces. Plea
do your scratch work on the backs of the exam pages. Write only **o**
answer to each problem; multiple answers will receive **no** credit, ev
one of them is correct.

Note: This examination runs for a total of 90 minutes. No questions
be answered by proctors after the exam begins. Please write legibly
ambiguous or messy answers will receive **no credit**.

(10 points) **Name** (using IUPAC nomenclature) or **draw**, as appropriate, the following molecules. Do not forget stereochemistry (e.g.: *cis*, *trans*, *R*, *S*, *meso*) where appropriate.

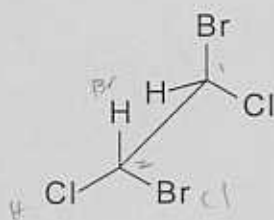
(a)



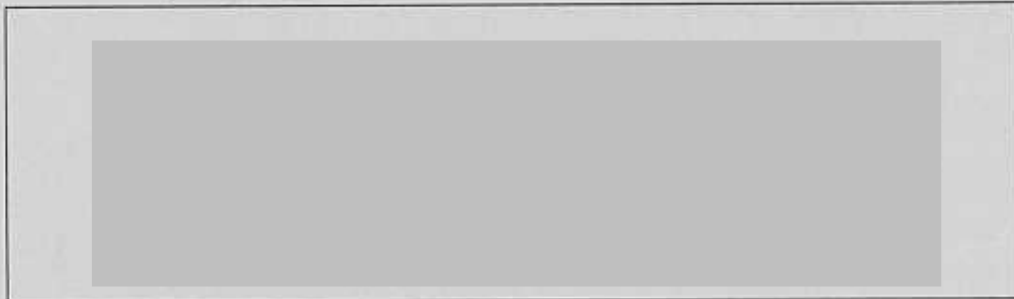
(b) (3R,4R)-3-bromo-4-methylhexane
(Show as a Fischer Projection)



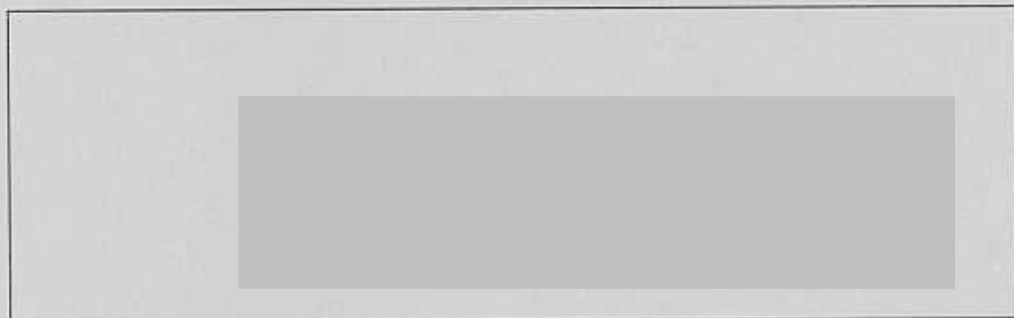
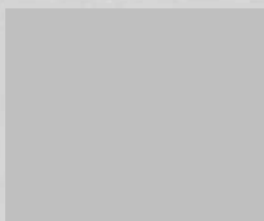
(c)



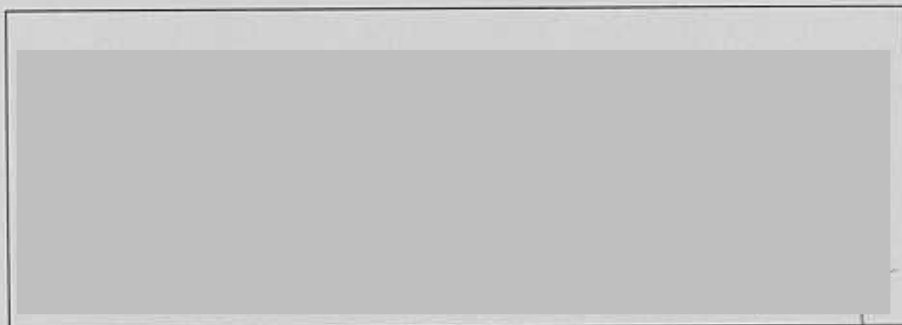
(Shown here as a sawhorse projection)



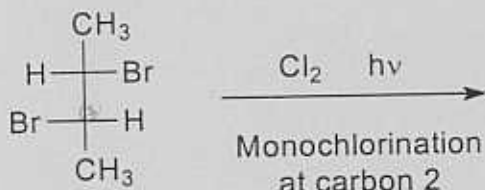
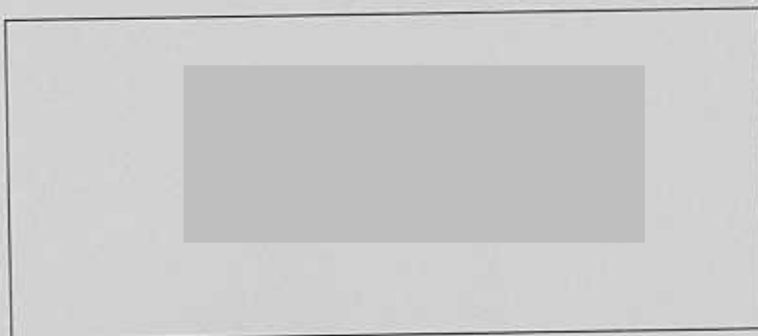
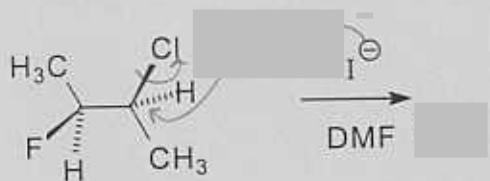
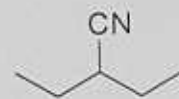
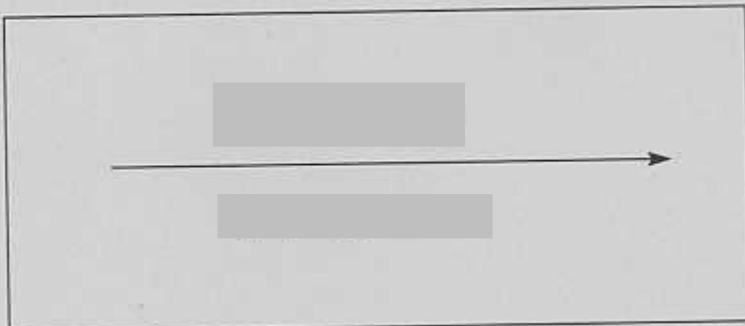
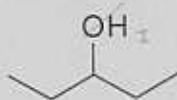
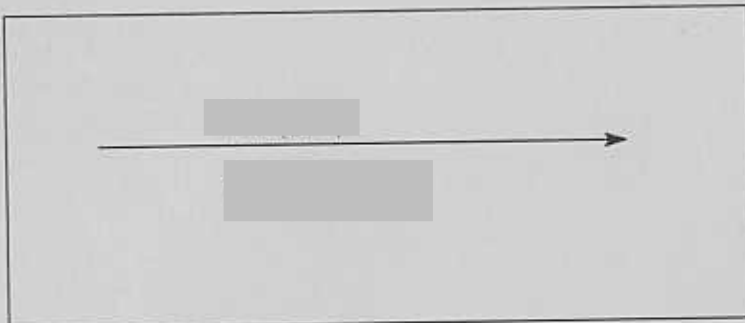
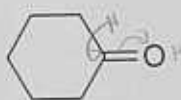
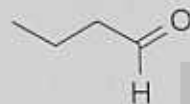
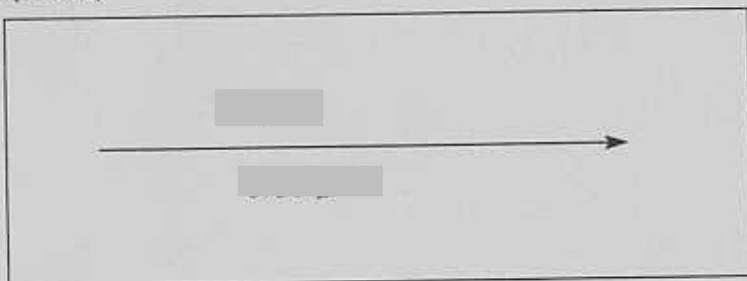
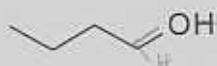
(d) (R)-6-fluoro-2-hexanol



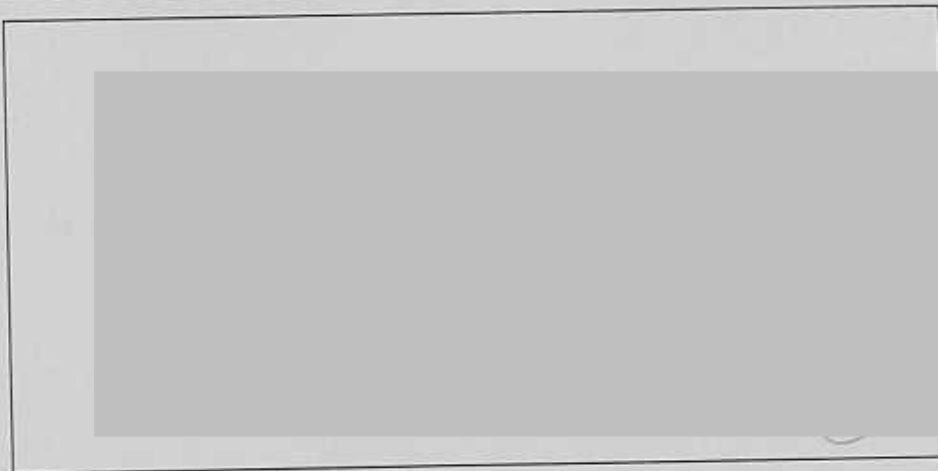
(e)



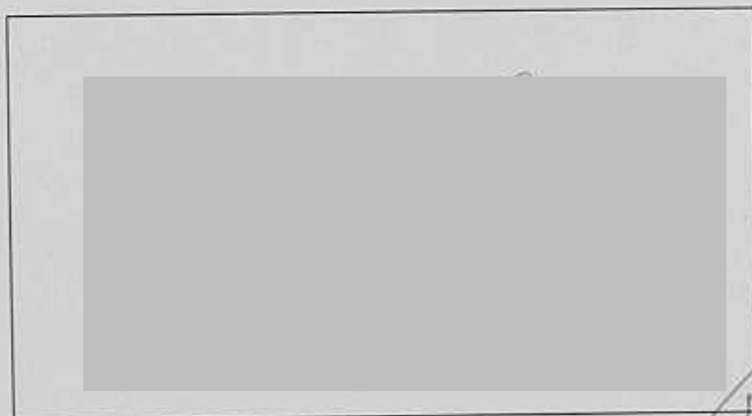
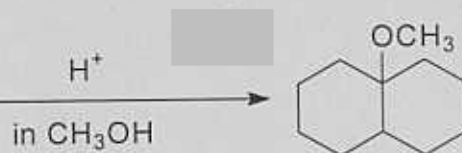
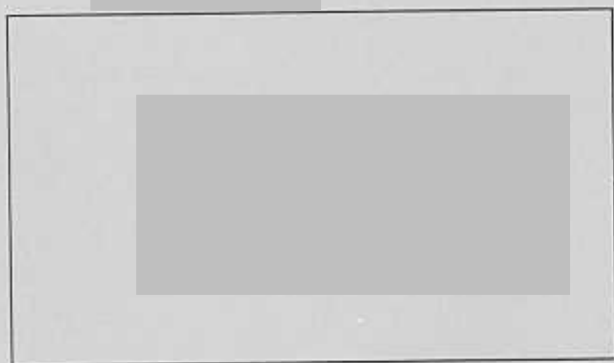
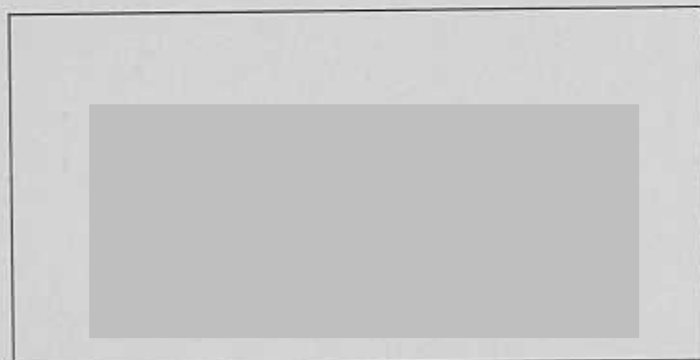
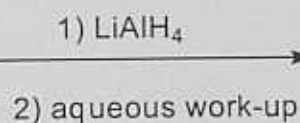
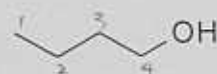
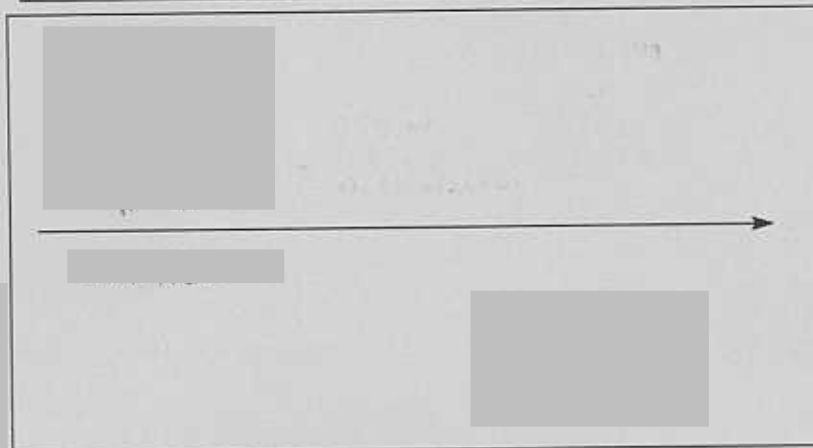
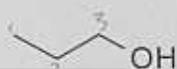
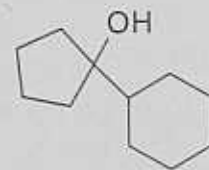
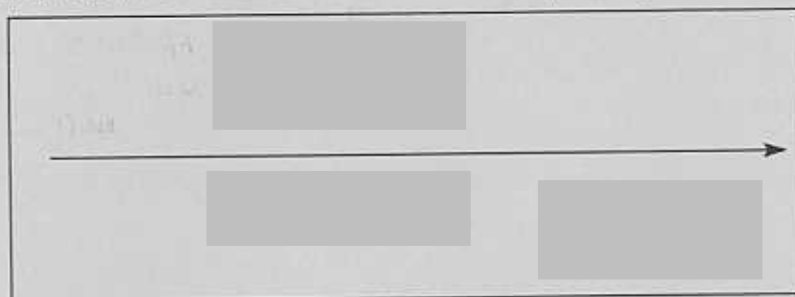
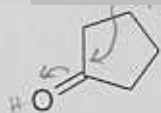
(11 Points). Add the missing reagents (and solvents, if relevant), or products (show stereochemistry where appropriate)



Hint: are there one or two products?

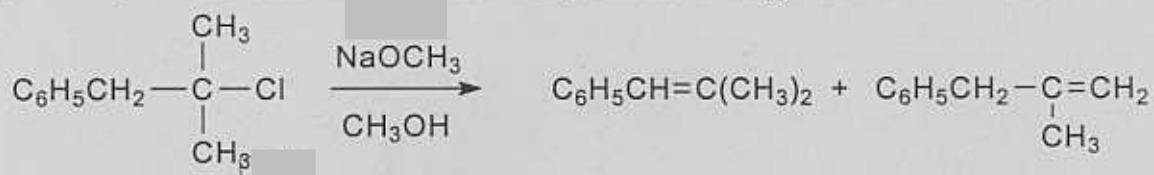


(12Points). Add the missing starting material, reagents, or products (show stereochemistry where appropriate, note that several reaction steps may be required!)



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4. (9 Points). The following reaction can proceed through both E1 and E2 mechanisms.



Data: the concentration of the haloalkane is 0.03M; the E1 rate constant is $k_1 = 1.5 \times 10^{-4} \text{ s}^{-1}$; the E2 rate constant is $k_2 = 1.8 \times 10^{-4} \text{ L mol}^{-1} \text{ s}^{-1}$.

- (i) Calculate the rate of each reaction and state which elimination mechanism is predominant with 1.5M NaOCH₃. Show detailed work and do not forget the units!

Rate of E1 reaction

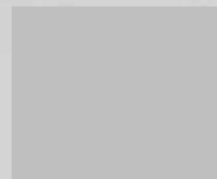
Rate of E2 reaction

Predominant mechanism is:

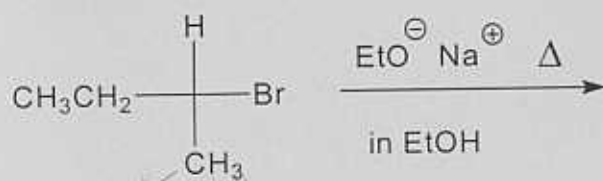
- (ii) At what concentration of base does exactly 50% of the starting material react by an E1 route and 50% by an E2 pathway? Show detailed work and do not forget units!

Answer:

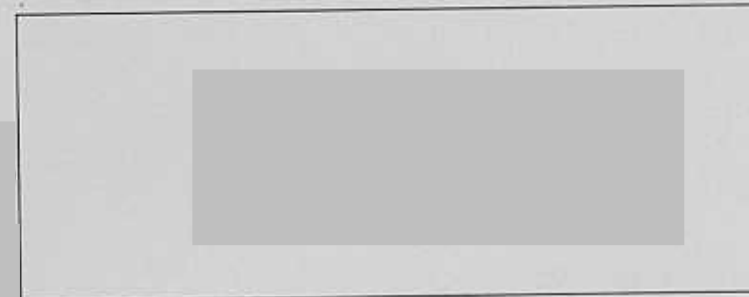
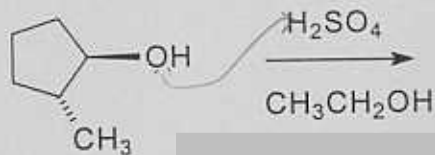
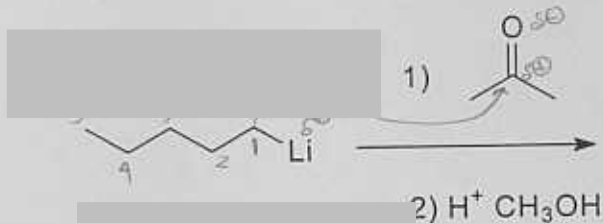
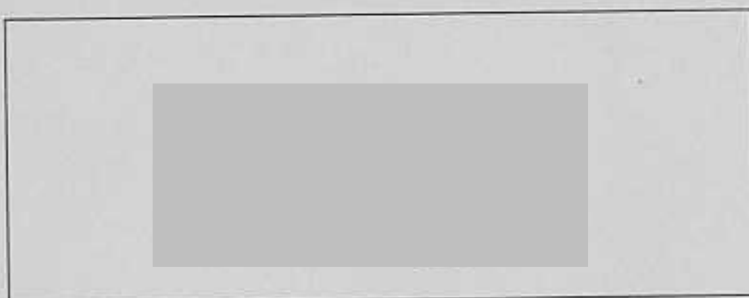
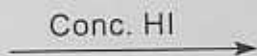
5. (11 Points). (a) Propose a synthesis of the (R) isomer of $\text{CH}_3\text{CHN}_3\text{CH}_2\text{CH}_3$ starting from (R)-2-chlorobutane.



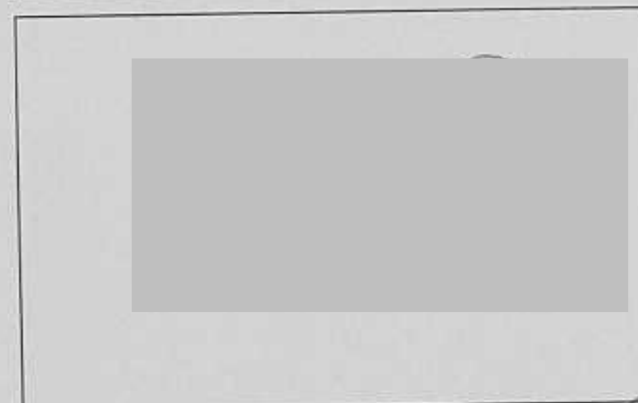
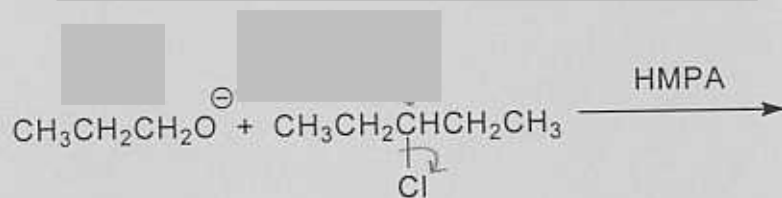
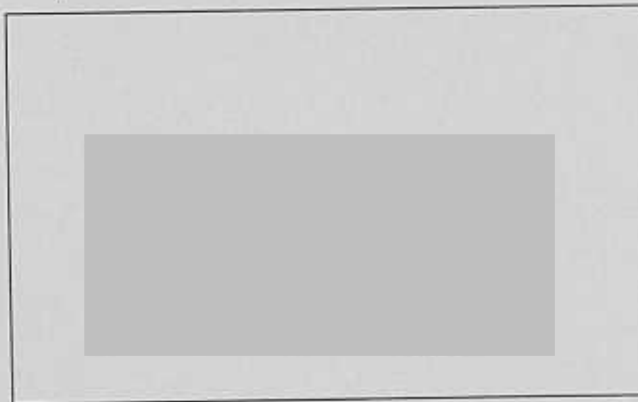
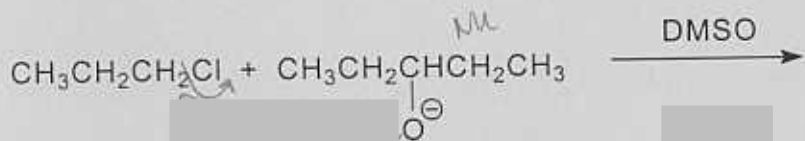
- (b) Show clear structures for **all** the products obtained in the following reaction and circle the major elimination product. (Note that there is no need to consider any "rearranged" product)



6. (11 Points). (a) Give the expected major product of each of the following reactions.



(b) Write the expected major product(s) of the following attempted ether syntheses

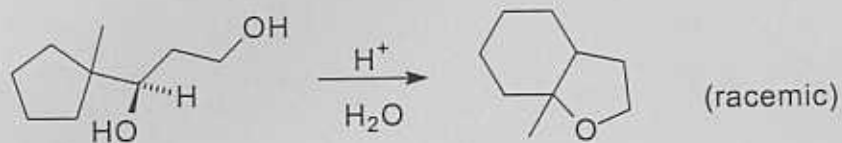


7. (14 points). Provide a viable synthetic route for the product below. You must start from the starting material indicated but you may use any other organic or inorganic compound in your synthetic scheme.



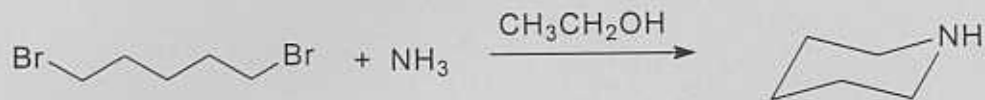
(b) Propose a high yield synthesis of 3-octanol starting from an aldehyde and an alkane

8. (10 points). Write a step-by-step mechanism (include arrows) for the reaction below. Show all intermediates, and explain (in the box) why the product is racemic.



Explanation: the product is racemic because...

9. (12 Points). (a) Show a detailed step by step mechanism (with curved arrows) explaining the outcome of the following reaction:



(b) Show clear structures for products A and B in the reaction sequence below:

