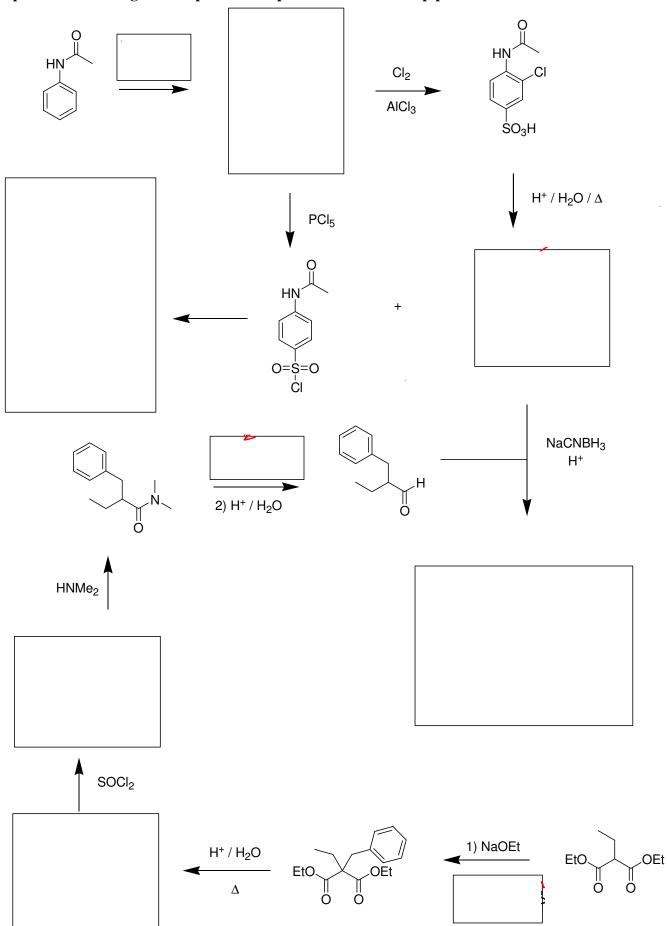
Name:	SII	D:	
Signature: PRINT YOUR NAME CLEARLY!!			
Chem 3B Su10 Neil O.L. Viernes	Midterm 2	- 	02AUG10
This exam has 11 pages; make su written on it will NOT be graded.	re you have them all. Page 11 is bl	lank. Use as scratch pape	er, anything
Please place answers in designated graded.	d spaces. Please write clearly. Mes	ssy or ambiguous answers	s will not be
This exam is 90 minutes long. No	clarifying questions will be answer	ed by the GSI's after the	exam begins.
Mark one of the following. If yo	ou are enrolled in Chem 3BL, mar	k off your laboratory se	ction.
Lecture Only	Completing I Grade		,
101 – Michael Chiang	(Professor Name)
102 – Amy McCarthy			
103 – Rob Padilla			
107 – Rob Padilla (Evening)			
108 – Kevin Zhao			
109 – Katherine He		Do not write	in this box
201 – David Nagle		1)	(12)
202 – Greg Dallinger		2)	(27)
203 – Reyu Sakakibara		3)	(16)
204 – Susan Kim		4)	(18)
207 – Arash Nayeri		5)	(20)
208 – Philip Chung		6)	(24)
		7)	(16)
		8)	
		6)	(15)
		9)	

1) (12 pts) Provide nomenclature or structures for the following:

Br NO ₂	
The state of the s	
N-Cyclohexylbutanamine	
Acetophenone	
Propanoic Anhydride	

2) (27 pts)

Complete the following roadmap. One compound or reaction step per box.



3) (16 pts)

Identify the most acidic hydrogen. Rationalize your answer.

Determine if the following molecules are aromatic, anti-aromatic or non-aromatic. Assume that all of these compounds cannot bend out of planarity.

Draw intermediate structures to help rationalize the substitution patterns observed for the following reactions.

$$\begin{array}{c|c} & & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & & \\ \\ & & \\ \hline \\ & &$$

4) (18 pts)

Provide the structures for the products for the following reactions.



Only one of the following trienes will undergo electrocyclic ring closing reactions. Circle your answer and provide an explanation.

Determine if heat or light is used to obtain the products specified. Identify the direction of the rotation (conrotary or disrotary).

Draw molecular orbitals rationalizing your answer to the first electrocyclic ring opening reaction above.

5) (20 pts) Provide a mechanism for the following transformation.

6) (24 pts) Provide a mechanism for the following transformation.

7) (16 pts)
Provide the best synthetic route to the following molecule.

From any mono-substituted benzene and any starting materials 6 carbons or less

8) (15 pts)

Provide the best synthetic route to the following molecule.

From and H₃CC=CCN (identify if the alkene is cis or trans) and any starting materials 3 carbons or less

9) (12 pts)

Provide the best synthetic route to the following molecule.