

Problem 1

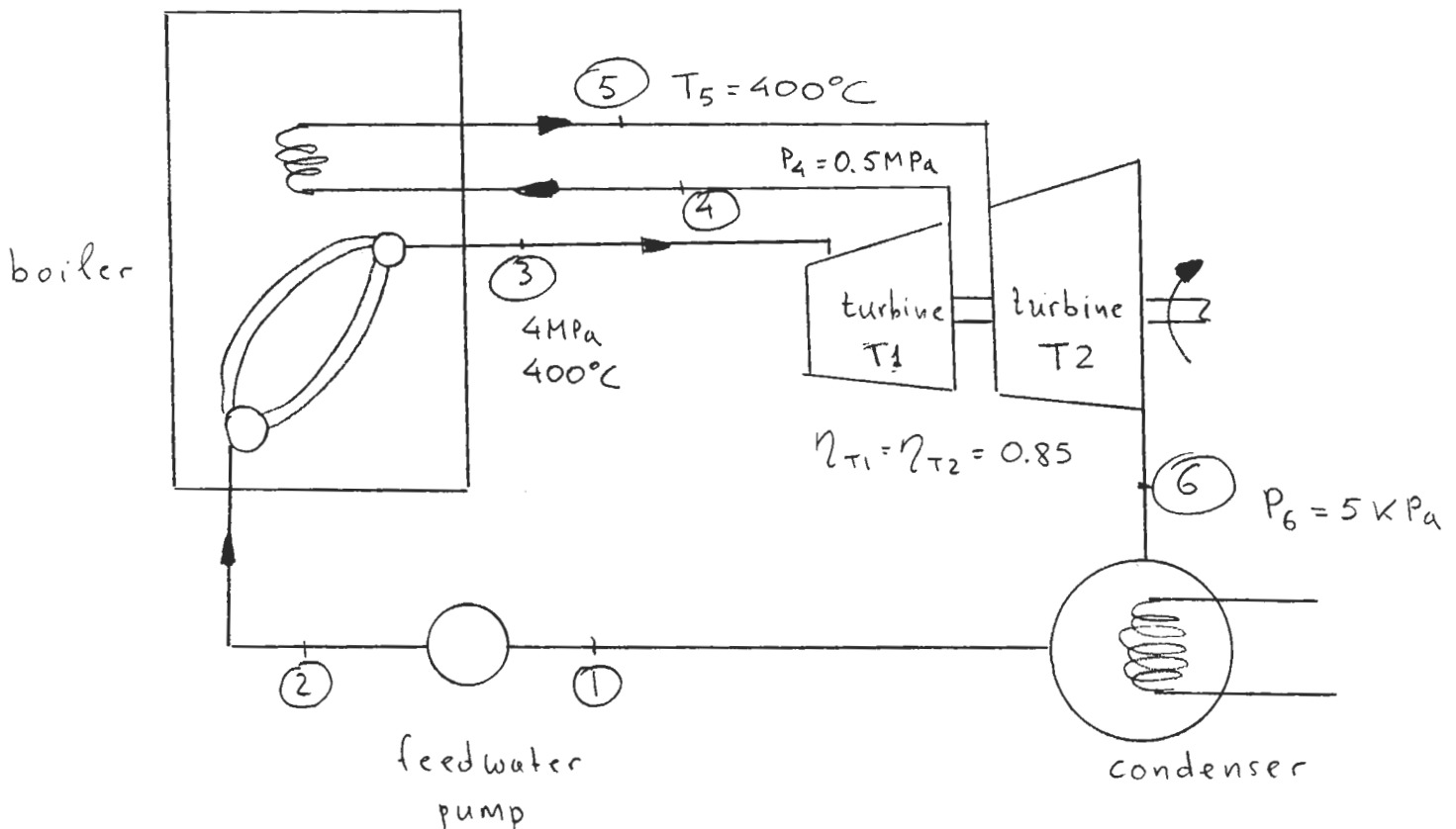
A well insulated piston-cylinder device contains 0.02 m^3 of saturated vapor refrigerant R-134a at a pressure of 1 MPa. The refrigerant is allowed to expand in a reversible manner till the pressure drops to 0.4 MPa. Please determine:

- The final temperature in the cylinder
- The work done on the refrigerant.

Problem 2

Consider a steam power plant that operates on a reheat Rankine cycle and has a net power output of 2 MW. Steam enters the turbine at 4 MPa and 400°C . The extraction to the reheat occurs at 0.5 MPa and the steam enters the second stage of the turbine at 400°C . The adiabatic efficiency of both stages of the turbine, $\eta_{T1} = \eta_{T2} = 0.85$. The condenser pressure is 5 kPa. Please:

- Show the cycle on T-s diagram
- Determine the mass flow rate of steam through the boiler
- Calculate the thermal efficiency of the cycle.



Problem 3

An air-standard cycle is executed in a closed system with 1 kg of air and consists of the following three processes:

- 1-2: Isentropic compression from 100 kPa and 20°C to 1 MPa
- 2-3: Isobaric (i.e. constant pressure) heat addition of 1000 kJ
- 3-1: Heat rejection to the initial state following the relation $P = c \cdot v$, where P is the pressure, v is the specific volume and c a constant.

Assuming that the specific heats C_p and C_v are constant, please:

- (a) Show the cycle on P - v and T - s diagrams
- (b) Calculate the cycle net work
- (c) Determine the cycle thermal efficiency.

Problem 1reversible + adiabatic \rightarrow isentropic $S_2 = S_1$

$$\begin{array}{l}
 P_1 = 1.0 \text{ MPa} \\
 \text{saturated vapor}
 \end{array}
 \left. \vphantom{\begin{array}{l} P_1 = 1.0 \text{ MPa} \\ \text{saturated vapor} \end{array}} \right\} \Rightarrow
 \begin{array}{l}
 v_1 = v_g @ 1 \text{ MPa} = 0.0202 \text{ m}^3/\text{kg} \\
 u_1 = u_g @ 1 \text{ MPa} = 247.77 \text{ kJ/kg} \\
 s_1 = s_g @ 1 \text{ MPa} = 0.9043 \text{ kJ/kgK}
 \end{array}$$

$$m = \frac{v}{v_1} = \frac{0.02}{0.0202} = 0.99 \text{ kg}$$

$$\begin{array}{l}
 P_2 = 0.4 \text{ MPa} \\
 S_2 = S_1
 \end{array}
 \left. \vphantom{\begin{array}{l} P_2 = 0.4 \text{ MPa} \\ S_2 = S_1 \end{array}} \right\} \Rightarrow x_2 = \frac{S_2 - s_f @ 0.4 \text{ MPa}}{s_g @ 0.4 \text{ MPa} - s_f @ 0.4 \text{ MPa}} \Rightarrow$$

$$x_2 = \frac{0.9043 - 0.2399}{0.9145 - 0.2399} = 0.985$$

$$T_2 = T_{\text{sat}} @ 0.4 \text{ MPa} = \underline{8.93^\circ \text{C}}$$

$$u_2 = u_f @ 0.4 \text{ MPa} + x_2 (u_g @ 0.4 \text{ MPa} - u_f @ 0.4 \text{ MPa}) \Rightarrow$$

$$u_2 = 61.69 + 0.985 (231.97 - 61.69) \Rightarrow$$

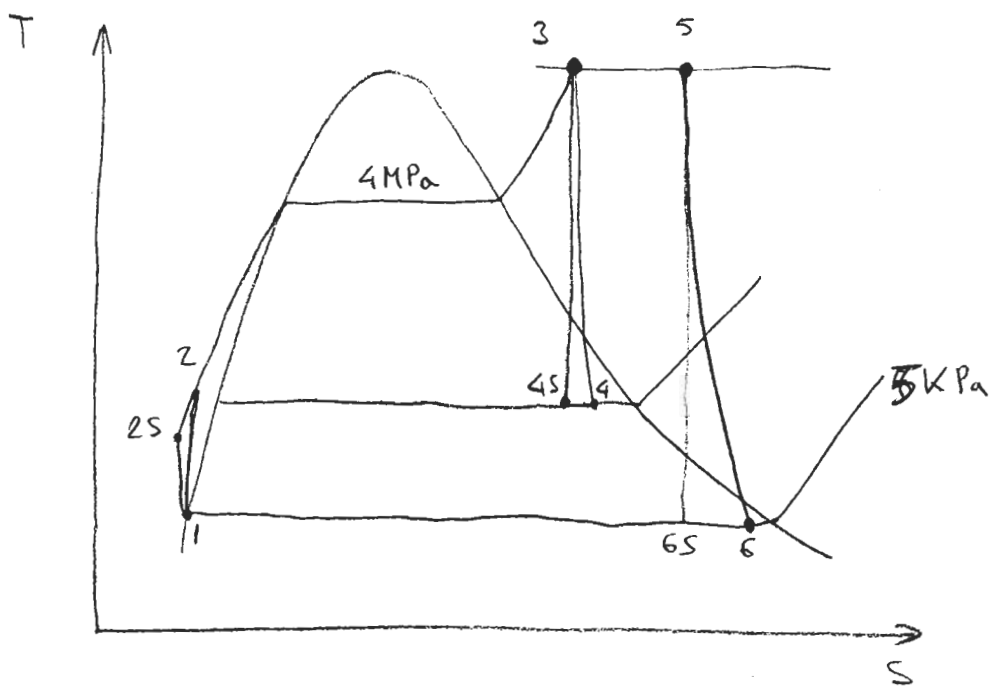
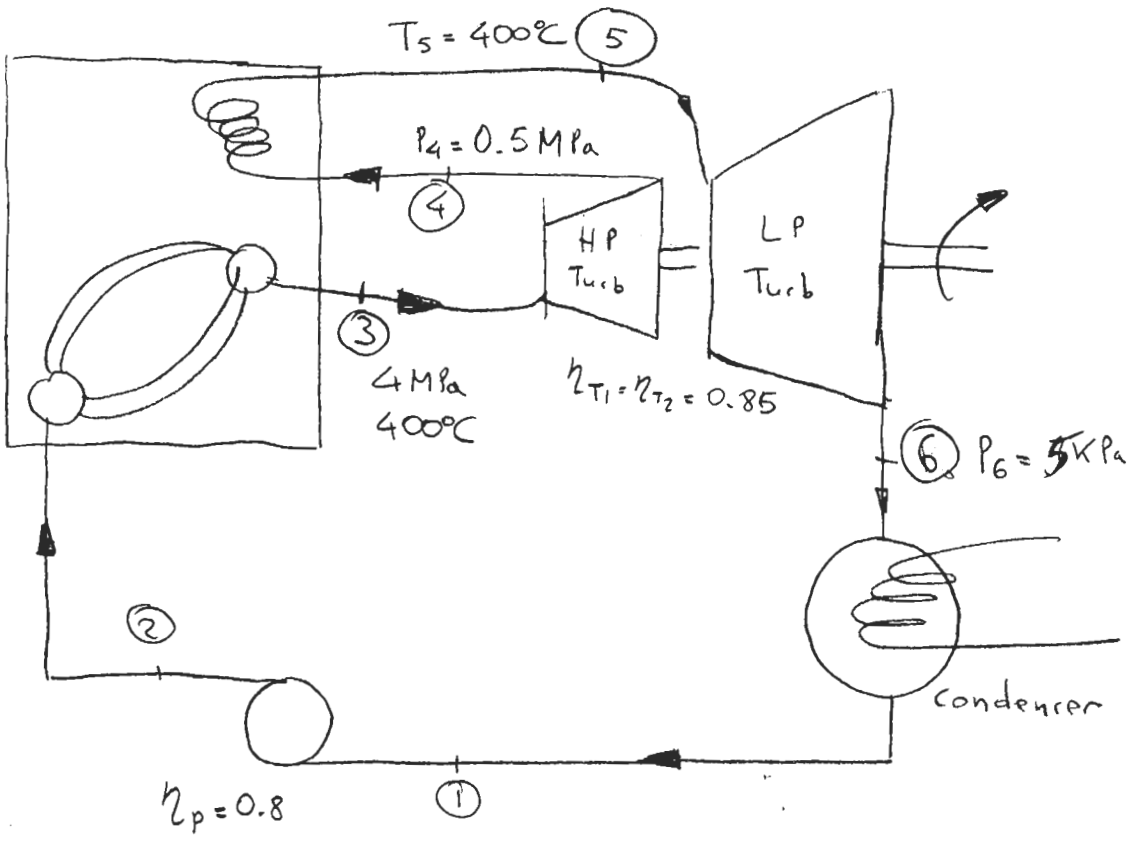
$$u_2 = 229.4 \text{ kJ/kg}$$

$$\cancel{Q} - \bar{W} = m(u_2 - u_1) \Rightarrow \bar{W} = -m(u_2 - u_1) \Rightarrow$$

$$\bar{W} = -0.99(229.4 - 247.77) \Rightarrow \bar{W} = 18.16 \text{ kJ}$$

Rankine Cycle with Reheat

$\dot{W}_{net} = 2 \text{ MW} \rightarrow$ (a) $\dot{m} = ?$
(b) $\eta_{th} = ?$



$$\left. \begin{matrix} P_3 = 4 \text{ MPa} \\ T_3 = 400^\circ\text{C} \end{matrix} \right\} \Rightarrow \begin{matrix} h_3 = 3213.6 \text{ kJ/kg} \\ s_3 = 6.7690 \text{ kJ/(kg K)} \end{matrix}$$

$$P_{4s} = 0.5 \text{ MPa} \Rightarrow \begin{matrix} s_f = 1.8607 \text{ kJ/(kg K)} \\ s_{fg} = 4.9606 \text{ kJ/(kg K)} \end{matrix}$$

$$x_{4s} = \frac{s_3 - s_f}{s_{fg}} \Rightarrow x_{4s} = \frac{6.7690 - 1.8607}{4.9606} = 0.99$$

$$h_{4s} = h_f + x_{4s} h_{fg} = 640.23 + 0.99 \cdot 2108.5 = 2727.62 \text{ kJ/kg}$$

$$h_3 - h_4 = \eta_{T1} \cdot (h_3 - h_{4s}) \Rightarrow h_4 = h_3 - \eta_{T1} (h_3 - h_{4s})$$

$$h_4 = 3213.6 - 0.85 (3213.6 - 2727.6) \Rightarrow$$

$$h_4 = 2800.5 \text{ kJ/kg}$$

$$\left. \begin{matrix} T_5 = 400^\circ\text{C} \\ P_5 = 0.5 \text{ MPa} \end{matrix} \right\} \Rightarrow \begin{matrix} h_5 = 3271.9 \text{ kJ/kg} \\ s_5 = 7.7938 \text{ kJ/(kg K)} \end{matrix}$$

$$P_{6s} = 5 \text{ kPa} \Rightarrow \begin{matrix} s_f = 0.4764 \text{ kJ/(kg K)} \\ s_{fg} = 7.9187 \text{ kJ/(kg K)} \end{matrix}$$

$$x_{6s} = \frac{s_5 - s_f}{s_{fg}} = \frac{7.7938 - 0.4764}{7.9187} = 0.924$$

$$h_{6s} = h_f + x_{6s} h_{fg} = 137.82 + 0.924 \cdot 2423.7 \Rightarrow$$

$$h_{6s} = 2377.3 \text{ kJ/kg}$$

$$h_5 - h_6 = \eta_{T2} \cdot (h_5 - h_{6s}) \Rightarrow$$

$$h_6 = h_5 - \eta_{T2} (h_5 - h_{6s}) \Rightarrow$$

$$h_6 = 3271.9 - 0.85 (3271.9 - 2377.3) \Rightarrow h_6 = 2511.5 \text{ kJ/kg}$$

$$W_T = W_{T1} + W_{T2} = (h_3 - h_4) + (h_5 - h_6) \Rightarrow$$

$$W_T = (3213.6 - 2800.5) + (3271.9 - 2511.5) \Rightarrow$$

$$W_T = 1173.5 \text{ kJ/kg} \quad \dot{m} = \frac{\dot{W}_T}{w_T} = \frac{2000}{1173.5} = 1.7 \text{ kg/s}$$

$$w_{sp} = v (P_2 - P_1) = 0.001005 \cdot (4000 - 5) = 4 \text{ kJ/kg}$$

$$w_p = \frac{w_{sp}}{\eta_{sp}} = \frac{4}{0.8} = 5 \text{ kJ/kg} \quad h_2 = h_1 + w_p = 137.82 + 5 = 142.82 \text{ kJ/kg}$$

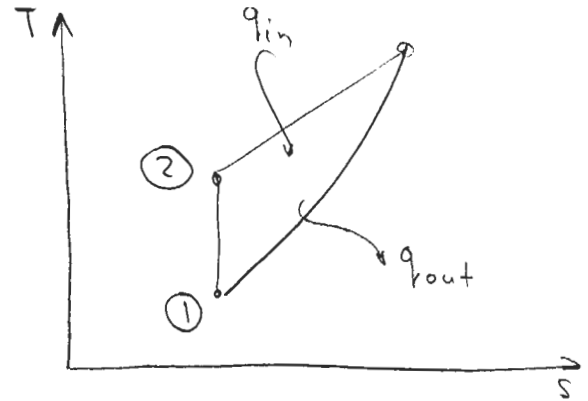
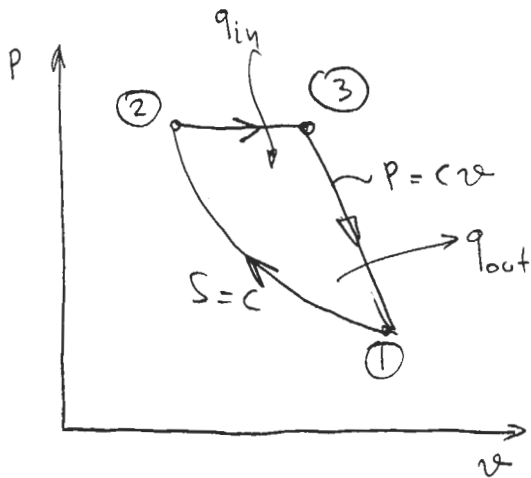
$$\dot{W}_p = \dot{m} w_p = 1.7 \cdot 5 = 8.5 \text{ kW}$$

$$q_H = (h_3 - h_2) + (h_5 - h_4) = (3213.6 - 142.82) + (3271.9 - 2800.5) \Rightarrow$$

$$q_H = 3542.18 \text{ kJ/kg}$$

$$W_{net} = W_T - W_p = 1173.5 - 5 = 1168.5 \text{ kJ/kg}$$

$$\eta_{TH} = \frac{W_{net}}{q_H} = \frac{1168.5}{3542.18} = 0.323$$

Problem 3

$$T_2 = T_1 \left(\frac{P_2}{P_1} \right)^{\frac{k-1}{k}} = 293 \left(\frac{1000}{100} \right)^{\frac{0.4}{1.4}} = 565.7 \text{ K}$$

$$Q_{in} = m(h_3 - h_2) = m c_p (T_3 - T_2) \Rightarrow$$

$$T_3 = \frac{Q_{in}}{m c_p} + T_2 \Rightarrow T_3 = \frac{1000}{1 \cdot 1.005} + 565.7 \Rightarrow T_3 = 1560 \text{ K}$$

$$w_{31} = \frac{P_3 + P_1}{2} (v_1 - v_3) = \frac{P_3 + P_1}{2} \left(\frac{R T_1}{P_1} - \frac{R T_3}{P_3} \right) \Rightarrow$$

$$w_{31} = \frac{1000 + 100}{2} \left(\frac{293}{100} - \frac{1560}{1000} \right) \cdot 0.287 \Rightarrow w_{31} = 216 \text{ kJ/kg}$$

$$Q_{31} - w_{31} = m(u_1 - u_3) \Rightarrow Q_{31} = m w_{31} + m c_v (T_1 - T_3)$$

$$Q_{31} = 1 \cdot [216 + 0.718 (293 - 1560)] \Rightarrow Q_{31} = -693.7 \text{ kJ}$$

$$\eta_{th} = 1 - \frac{Q_{out}}{Q_{in}} = 1 - \frac{|Q_{31}|}{Q_{in}} = 1 - \frac{693.7}{1000} \Rightarrow$$

$$\eta_{th} = 0.3063$$

$$W_{net} = Q_{in} - Q_{out} \Rightarrow W_{net} = 1000 - 693.7 = 306.3 \text{ kJ}$$