

Name: GSI Key

SID: _____

Signature: _____

**PRINT YOUR
NAME CLEARLY!!**

**Chem 3B Su10
Neil O.L. Viernes**

Midterm 2

02AUG10

This exam has 11 pages; **make sure you have them all.** Page 11 is blank. Use as scratch paper, anything written on it will NOT be graded.

Please place answers in designated spaces. **Please write clearly.** Messy or ambiguous answers will not be graded.

This exam is 90 minutes long. No clarifying questions will be answered by the GSI's after the exam begins.

Mark one of the following. If you are enrolled in Chem 3BL, mark off your laboratory section.

Lecture Only

Completing I Grade
(Professor Name _____)

101 – Michael Chiang

102 – Amy McCarthy

103 – Rob Padilla

107 – Rob Padilla (Evening)

108 – Kevin Zhao

109 – Katherine He

201 – David Nagle

202 – Greg Dallinger

203 – Reyu Sakakibara

204 – Susan Kim

207 – Arash Nayeri

208 – Philip Chung

Do not write in this box

1) _____ (12)

2) _____ (27)

3) _____ (16)

4) _____ (18)

5) _____ (20)

6) _____ (24)

7) _____ (16)

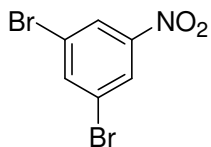
8) _____ (15)

9) _____ (12)

Total: _____ (160)

1) (12 pts)

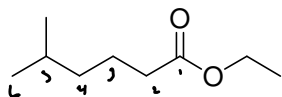
Provide nomenclature or structures for the following:



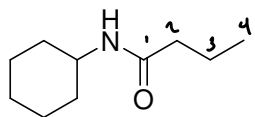
1,3-Dibromo-5-nitrobenzene

or

3,5-Dibromonitrobenzene

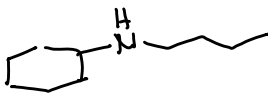


Ethyl-5-methylhexanoate



N-Cyclohexylbutanamide

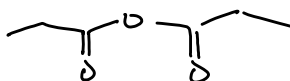
N-Cyclohexylbutanamine



Acetophenone

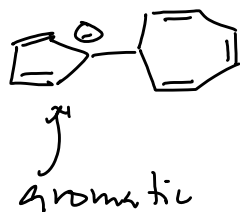
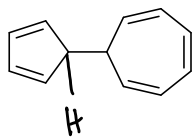


Propanoic Anhydride

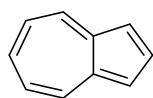


3) (16 pts)

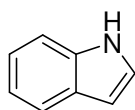
Identify the most acidic hydrogen. Rationalize your answer.



Determine if the following molecules are aromatic, anti-aromatic or non-aromatic. Assume that all of these compounds cannot bend out of planarity.



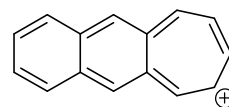
Aromatic



Aromatic

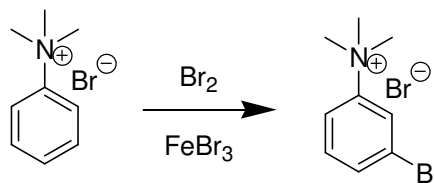
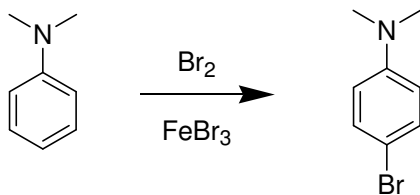


Anti-aromatic



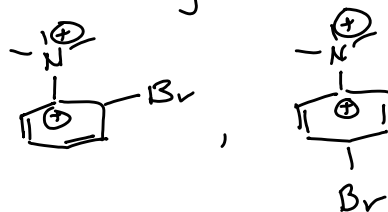
Aromatic

Draw intermediate structures to help rationalize the substitution patterns observed for the following reactions.

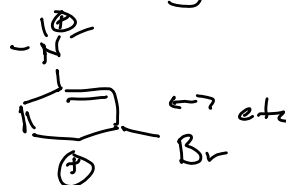


Additional resonance structure stabilizes the para-substituted intermediate

ortho/para subst. would generate a destabilizing resonance contributor

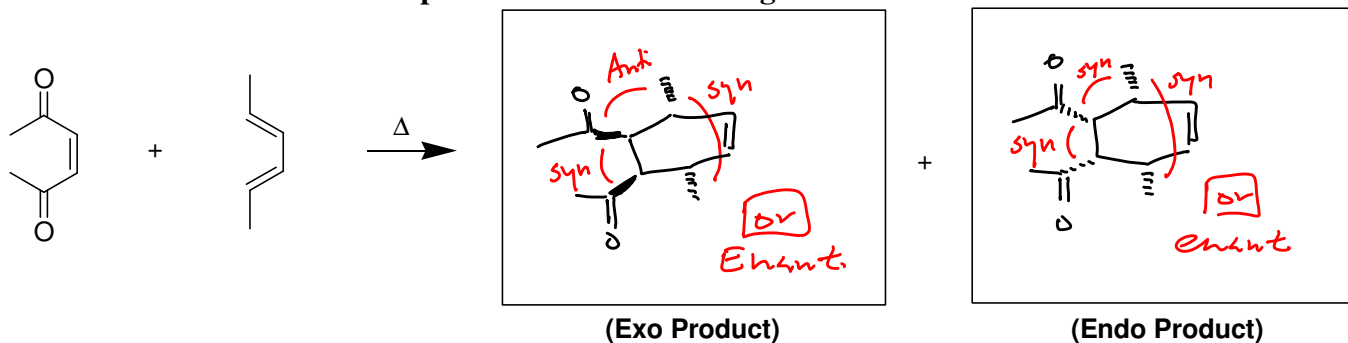


meta substitution does not have this destabilizing contributor

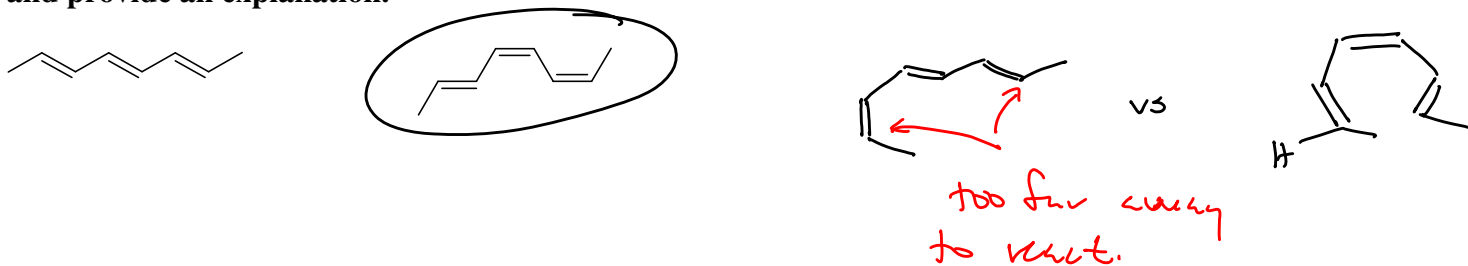


4) (18 pts)

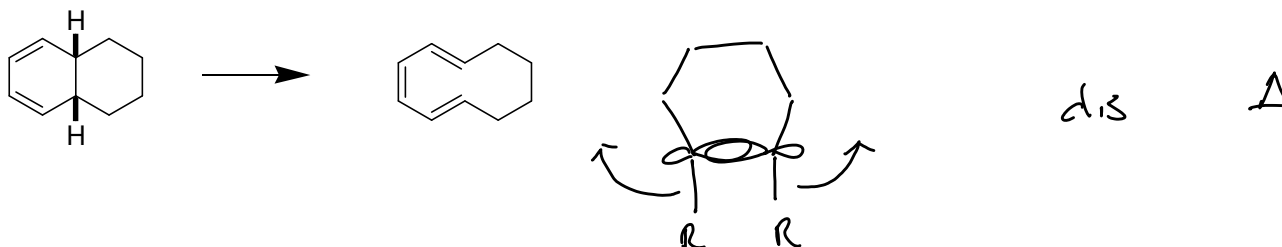
Provide the structures for the products for the following reactions.



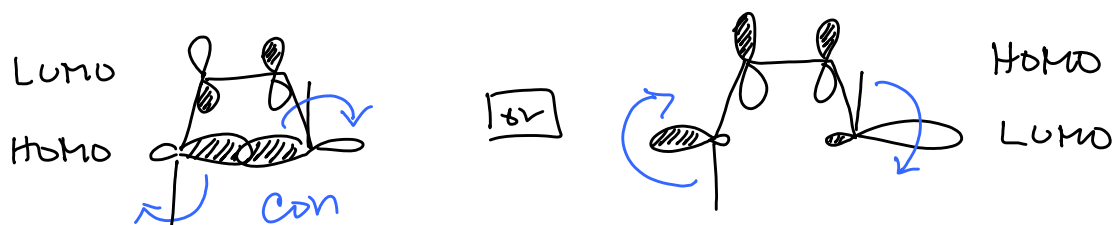
Only one of the following trienes will undergo electrocyclic ring closing reactions. Circle your answer and provide an explanation.



Determine if heat or light is used to obtain the products specified. Identify the direction of the rotation (conrotatory or disrotatory).

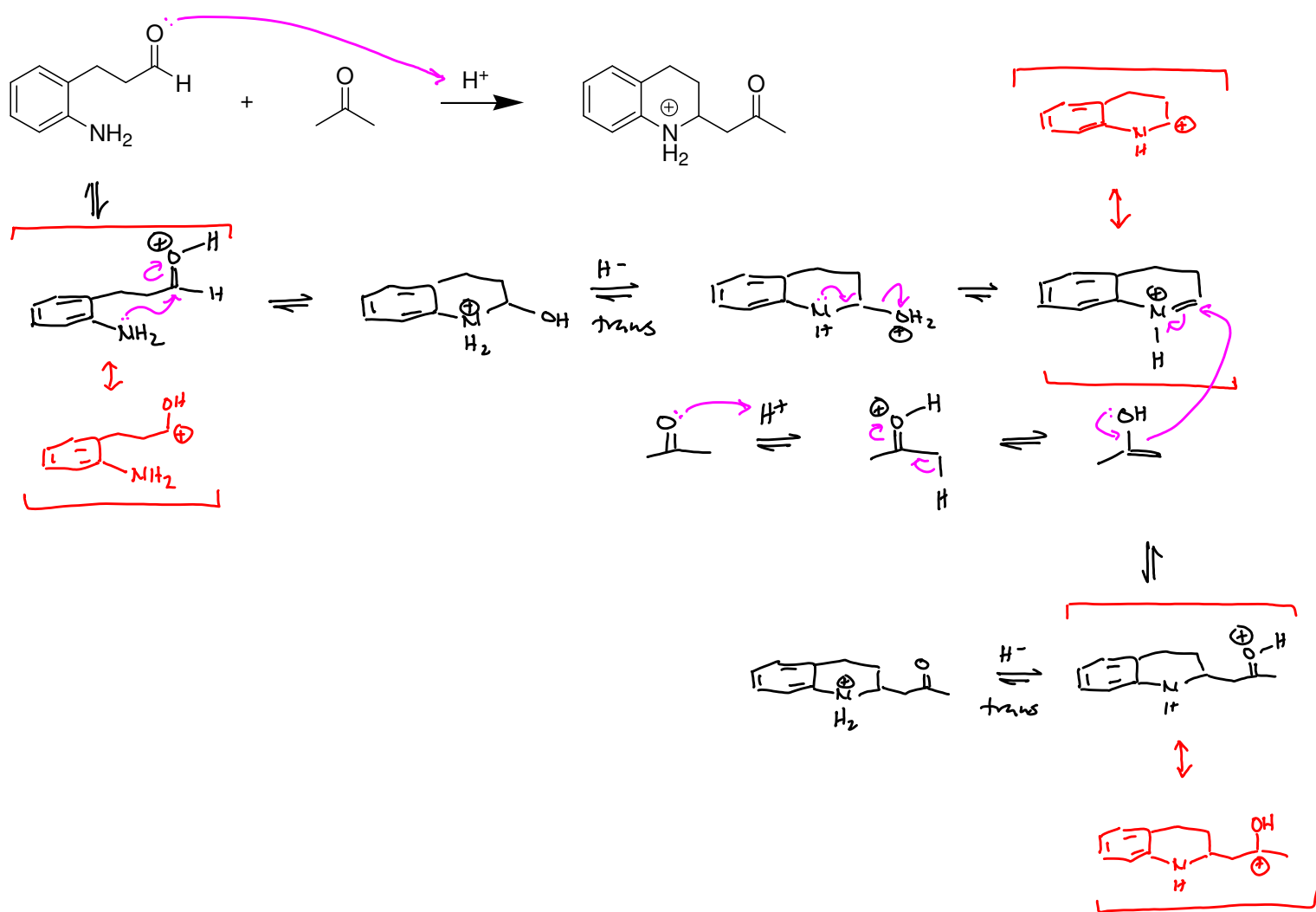


Draw molecular orbitals rationalizing your answer to the first electrocyclic ring opening reaction above.



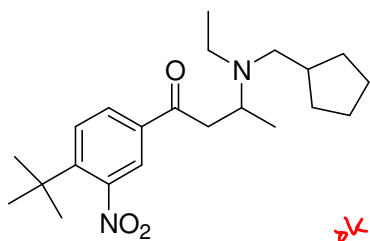
6) (24 pts)

Provide a mechanism for the following transformation.

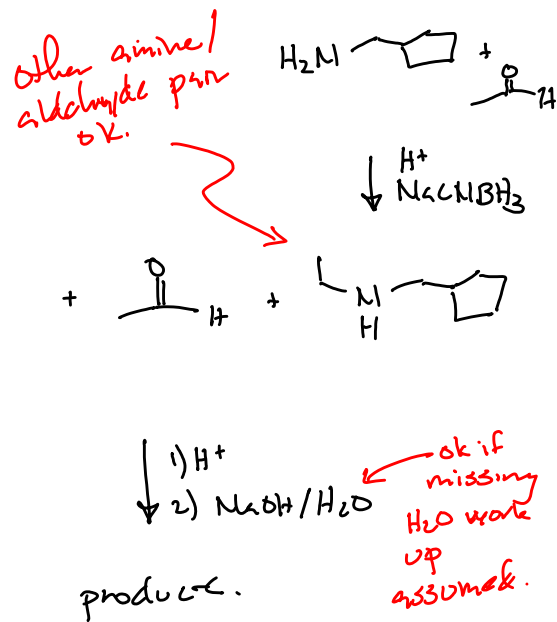
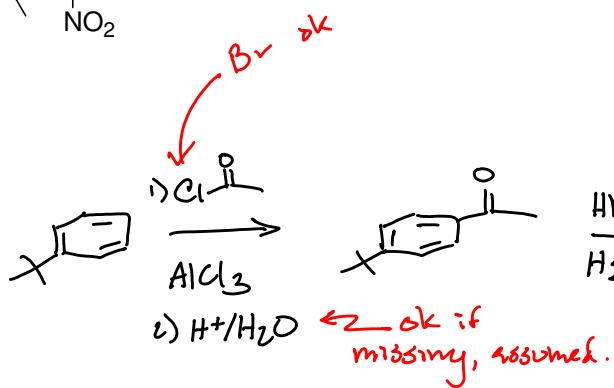


7) (16 pts)

Provide the best synthetic route to the following molecule.

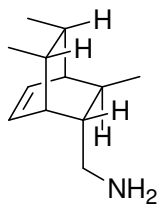


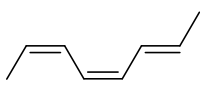
From any mono-substituted benzene and any starting materials 6 carbons or less

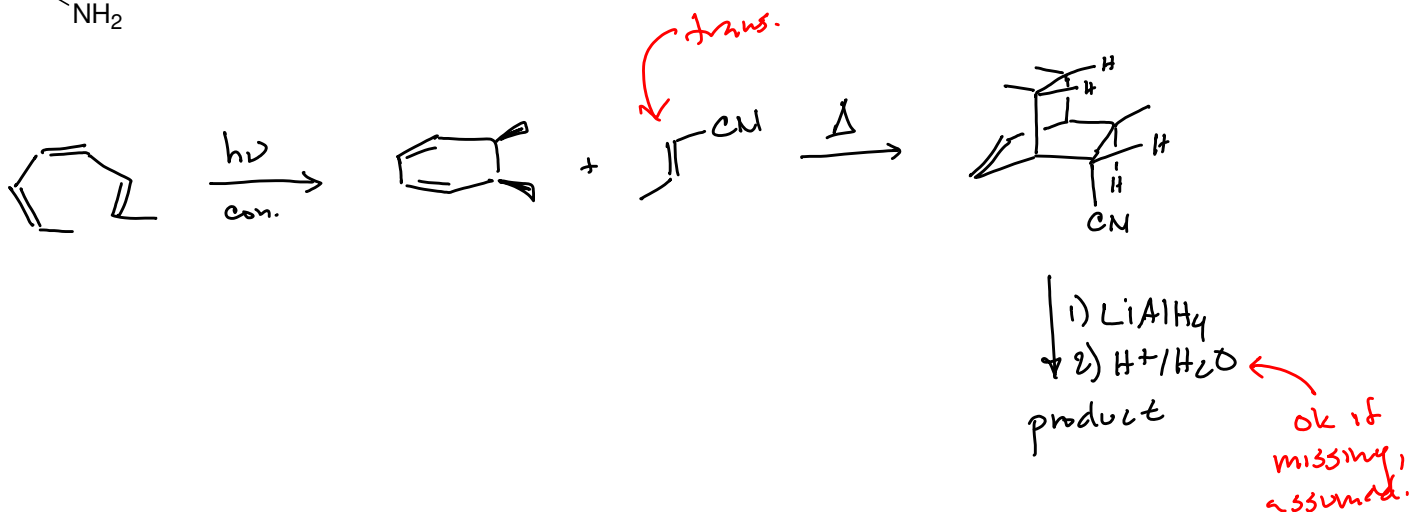


8) (15 pts)

Provide the best synthetic route to the following molecule.



From  and $\text{H}_3\text{C}-\overset{\text{H}}{\text{C}}=\overset{\text{H}}{\text{C}}-\text{CN}$ (identify if the alkene is cis or trans) and any starting materials 3 carbons or less



9) (12 pts)

Provide the best synthetic route to the following molecule.

