

PHYSICS 7B – Fall 2010
Midterm 1, R. Ramesh
Monday, September 27, 2010

Use the convention that $\Delta E = Q - W$ on this exam.

Problem 1 (20 points)

Consider a gas being blown along at a velocity $\mathbf{u} = u\hat{z}$, so that its velocity distribution is given by

$$F(\mathbf{v}) = \frac{1}{Z} e^{-m(\mathbf{v}-\mathbf{u})^2 / 2kT}.$$

Note that this is a probability distribution for the vector quantity \mathbf{v} , not the scalar speed $v=|\mathbf{v}|$, and has units of [velocity]⁻³.

- Find $\langle \mathbf{v} \rangle$, $\langle v^2 \rangle$ and v_{rms} .
- Find the peak velocity where $F(\mathbf{v})$ is maximized.

Problem 2 (15 points)

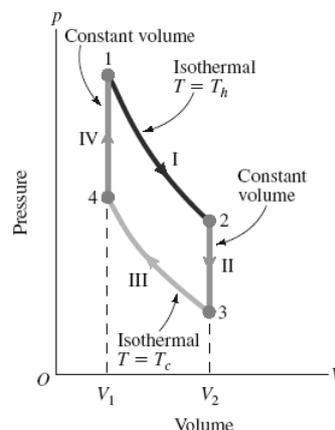
One hundred grams of ice at 0°C is dropped into 200g of water at 49°C. The system is thermally isolated. After a period of time, the ice has entirely melted, leaving 300g of water at 6°C. Assume the specific heat of water is constant and equal to 1JK⁻¹kg⁻¹.

- Calculate the latent heat of fusion for water.
- Calculate ΔS for the entire system.

Problem 3 (25 points)

For the thermodynamic cycle on the right with an ideal diatomic gas as the working material,

- Calculate W and Q for each of the four sides of the PV diagram.
- Sketch T vs. S for this process. You need not indicate specific values of T or S on your plot, but label the points 1-4 corresponding to those on the P-V diagram.
- Compare the efficiency of this engine with the efficiency of a Carnot engine for $T_H=400$ K, $T_C=300$ K, $V_1=1$ L, and $V_2=5$ L.



Problem 4 (15 points)

Using what you know about heat conduction, derive equations for the effective thermal conductivity of two materials with the same area and thickness but different thermal conductivities k_1 and k_2 when

- The materials have are arranged in series (heat flows through one then through the other).
- The materials conduct heat in parallel (heat flows through both simultaneously).

Problem 5 (25 points)

Use a combination of heat engines and heat pumps to prove that no engine can be more efficient than a Carnot engine when operating between a given maximum and minimum temperature.