March 18, 2009

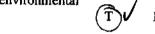
## **MIDTERM**

ascimento

Provide support for your answers. Be organized, concise and neat. Calculations must be clearly presented to obtain partial credit. (100 points)

Part A: TRUE/ FALSE QUESTIONS	
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- 1. The double layer of a clay particle, t, will decrease with decreasing dielectric constant, with increasing cation concentration and increasing valence of exchangeable cation.
- 2. To obtain the lowest hydraulic conductivity for a compacted clay liner you must compact the soil with a water content higher than the optimum water content (wet side of the line of optima) in order to obtain a "dispersed structure"
- 3. In general, as you decrease the hydraulic conductivity of compacted clay liners, the main mechanism of contaminant migration changes from advection-dispersion to constrained molecular diffusion
- 4. A material is considered hazardous when it contains a material listed as hazardous by the USEPA, displays hazardous characteristics (i.e., ignitability, corrosivity, reactivity or toxicity) and/or causes environmental or health damage



(2.5 pt each, total=10 pts)

Part B: Respond briefly to the following questions
 Describe (very briefly) some of the difficulties encountered in determining waste properties such as unit weight, strength and compressibility.

· lack of date due to heteroxeneas nature of waste, societal behavior · huste varies in different locations · difficult to test

- List (at least three) processes affecting the movement and change in concentration of a contaminant in groundwater and how would you reduce the risk associated with those processes in a waste containment system
- · adrection + mechanical dispersion -> minimize k to reduce exercise relating
- · diffusion -> force a minimuph thickness of material/Inner

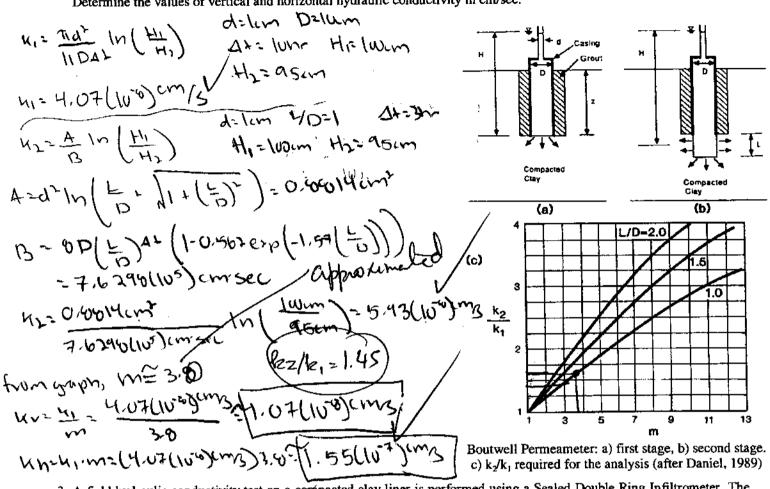
· absorptions adourption forgineer lines to about condaminants or destroy them (i.e. readire borrier)

 Describe briefly what is chemical compatibility testing and what is the main testing difference between low hydraulic conductivity soils (e.g., clays) vs. high hydraulic conductivity soils (e.g., sands).

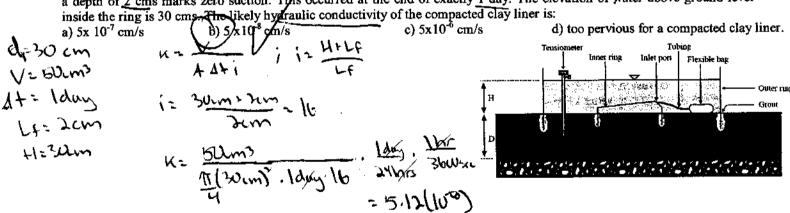
chamital compatibility testing is performing a test for nater than compare/use to test of containments expected in the freed in order to assess long term a volume to use when testing is when it of soil stabilities are time. For clay, I his is verally 2-4 pure textines for sands, it ispless so we immersely sands in leachate for 2-3 months and then test them

_	,		ture for hydraulic barrier	applications
, add fre	s (clay) by 6	phmiz du	1 density	
, 299 WIM	r amount uf/b	entonite t	o meet K reading	ements
o Orchiona	the same of the same			expected in the
field i	n order to ass	iss long &	em K.	
			,	
5. List three key clays.	elements of the Sealed	Double Ring Infiltr	ometer used for field test	ing of hydraulic conductivity for
+ measo	volume of exil	·	nat	what Dasked
* Inch	volume at spil	is permed	460 1001	001000
* finds	win the vo	utical dia	echion	
	V			
			. 1	
6. List one advar	ntage and one disadvanta	ge for laboratory h	ydraulic conductivity tes	ting for compacted clays
	ntage of Laboratory Te		Disadvanta	ge of Laboratory Testing
bandary co	ng your are an	10mm/	application of	reals to real silvuttin
•	`	V	is but alway	s actuale
1. A 0.60m thick project. This mate inquire on several	drainage blanket construerial is not available in geonets available in the p-plane flow. The flow	icted with gravely sufficient quantitie market. As a result	s (at reasonable cost) and, the following data is gi	specified in the design phase of your d you suggested to your supervisor to wen to you to assess the equivalency of acts at a confining pressure of 100 kPa
Geonet number	Name	Ma	nufacturer	Transmissivity (m³/sec per meter)
1	Enkanet 4015	Akzo Nobel	Geosynthetics Co.	0.039 🗙 0.7-
2	CE 3		rax Corp.	6.0 × 810.0
3	ТептаNet 200		BTEC/Inc.	0.026 10.7
analysis, the g a. only #1 الایمار که در که	eonets satisfying or exce b. onl	eding the performa y # 1 and # 3	nthetic product is expense of the original drains c. all of them  M/S = 0.05 M/S  Over home  1 100 + 1	d. note of them 10052060
amage =>	, hart large !	47		
Mune	1 1	are largor	than Wand S	It same Michnelsof
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2. A field hydraulic conductivity test is performed in a compacted clay liner with a Boutwell permeameter, with dimensions, d= 1 cm, D = 10 cms. For the initial stage (a), the initial height H was 100 cm, and the head drop in 10 hr was 5 cm. For the second stage (b), and a factor L/D=1 a head drop of 5 cm was observed in 3 hrs, from an initial height of 100 cm. Determine the values of vertical and horizontal hydraulic conductivity in cm/sec.



3. A field hydraulic conductivity test on a compacted clay liner is performed using a Sealed Double Ring Infiltrometer. The inner ring is 30 cms in diameter and a volume of 50 cm<sup>3</sup> is measured to have gone into the soil at the time the tensiometer at a depth of 2 cms marks zero suction. This occurred at the end of exactly 1 day. The elevation of water above ground level



Bonus Question: Identify the following geosynthetic specimen, its primary function in a waste containment system and main properties (do not list all properties, but the 1-2 most relevant) (2 pt/each)

Specimen	Geosynthetic Type	Potential Functions	Main Properties of Interest
	oso dug	remfuremens	Shong in known when force appliced along primary axis

	<u> </u>
Table for	The Equation for 1-D transport with Advection, Diffusion/Dispersion & Retardation is given by:
-	The Equation for 1-D naisport with Advection, Diffusion/Dispersion & Retardation is given by:
Complementary	C 1 ( D = V 1)
Error Function	$\frac{C}{A} = \frac{1}{2} \int_{A} \frac{R \cdot Z - V_S \cdot V_S}{A \cdot Z - V_S \cdot V_S} \left( \frac{V_S \cdot Z}{A \cdot Z} \right) = \frac{1}{2} \left( \frac{R \cdot Z + V_S \cdot V_S}{A \cdot Z} \right)$
B enfc(B)	$\frac{C}{C_0} = \frac{1}{2} \left\{ \operatorname{erfc} \left( \frac{R.z - V_s t}{2\sqrt{D.R.t}} \right) + \exp \left( \frac{V_s. z}{D} \right) \operatorname{erfc} \left( \frac{R.z + V_s t}{2\sqrt{D.R.t}} \right) \right\}$
	ZID.R.t /)
	For the annual and a second and
0.05 0.94363	For the case of no seepage velocity $(V_{\bullet} = I)$ , the previous equation simplifies to:
	For the case of no seepage velocity $(V_s = 0)$ , the previous equation simplifies to:
	$\Delta = arfo(-K,Z)$
0.15 0.83200	$\frac{C}{C_0} = \text{erfc}\left(\frac{R.z}{2\sqrt{D.R+}}\right)$
0.20 0.77730	1 Co \2\D.R.t /
0.25 0.72367	1. 4. A clay liner 90 cms (~ 3ft) thick with a hydraulic conductivity of 10-8
0.30 0.67137	4. A clay liner 90 cms (~ 3ft) thick with a hydraulic conductivity of 10 s cm/s and an effective porosity of
	0.333 is constructed at the base of a landfill. Assuming that the manifest that the
0.35 0.62062	0.333 is constructed at the base of a landfill. Assuming that the maximum elevation of the leachate at the
0.40 0.57161	top of the liner would be only 30 cm (~1ft) and the pore pressure below the liner is assumed as a state of the liner is a state of the li
	top of the liner would be only 30 cm (~1ft) and the pore pressure below the liner is assumed atmospheric, the time required for the leachate to appear at the bottom of the liner considering only advection would be most
0.45 0.52452	time required for the leachate to appear at the bottom of the liner considering only advection would be most
0.50 0.47950	likely:
	inkery.
0.55 0.43668	a. ~ 7 years b. ~ 14 years c ~ 25 years
0.60 0.39614	a. ~ 7 years b. ~ 14 years c. ~ 25 years d. 70 years
0.65 0.35797	
	2:90m unhy advection
0.70 0.32220	Oring advection
0.75 0.28884	1 150 m/
	4. 100 m/s sque
0.80 0.25790	$\gamma$
0.85 0.22933	10 (123= X Cur
	1 1/20/333
0.90 0.20309	2=90m unity advection vs. \(\frac{1}{2}\) = \(\frac{1}\) = \(\frac{1}{2}\) = \(\frac{1}{2}\) = \(\frac
0.95 0.17911	1 Ah - 4am Die
1.00 0.15730	- The state of the
	I will the start of the start o
1.05 0.13756	->MICONOMICONA NO
1.10 0.11979	2 2
1	5 The commentation of a mariable to the state of the stat
	5. The concentration of a particular chemical in the leachate of a landfill is 1000 ppb (parts per billion). The
1.20 0.08969	constrained molecular diffusion for this chemical is D*= 5 x 10 <sup>6</sup> cm <sup>2</sup> /sec. Assume for this part of the
1.25 0.07710	constrained inforcement diffusion for this chemical is $0^{\pm} = 5$ y $10^{\circ}$ cm <sup>-leac</sup> . Accomp for this next of the
	problem that the contaminant is not adopted into the time of the part of the
1.30 0.06599	problem that the contaminant is not adsorbed into the liner (R=1). If the initial background concentration is
1.35 0.05624	zero and that transport results only from molecular diffusion, the concentration of this chemical at the bottom
1.00 0.00024	and that transport estats that from indicediar diffusion, the concentration of this chemical at the bottom
1.40 0.04772	of a clay liner (thickness ~ 90 cms) at the end of 5 years is approximately:
1.45 0.04031	so cans at the cha of y scale is approximately:
1.50 0.03390	(8, 20-25 ppk) h 200-250 pph
1.50 0.03390	1
1.55 0.02838	L Carlotto
1.60 0.02365	1000
1.00 0.02303	1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -
1.65 0.01962	D=25(1) \ XC \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \
1.70 0.01621	(240HQ)
1.75 0.01333	1 2 1 2 1 2 1 2 1 2 1 2 1 2 1 2 1 2 1 2
1.75 0.01333	1 \\2\
1.80 0.01091	
1.80 0.01091	7-91km = 1147 orfol/1603/ 2/41/000000 1/00/2/2 2261
1.85 0.00889	= 90m = 100 erfel 1.603) = 100,0,0,0365 / 200 = 23,65
1.85 0.00889 1.90 0.00721	= 90m = 100 erfel 1.603) = 100.0.03365 / 200 = 23,65
1.80 0.01091 1.85 0.00889 1.90 0.00721 1.95 0.00582	= 90m = 100 erfe(1.603) = 100.0.03365 / 200 = 23,63
1.80 0.01091 1.85 0.00889 1.90 0.00721 1.95 0.00582	2 = 90m = 100 erfe(1.603) = 100.0000 / 200 = 23,65
1.80 0.01091 1.85 0.00889 1.90 0.00721 1.95 0.00582 2.00 0.00468	2 = 90m = 100 erfel 1.603) = 100.0.03365 1/200 = 23,65
1.80 0.01091 1.85 0.00889 1.90 0.00721 1.95 0.00582 2.00 0.00468 2.05 0.00374	= 100 erfe(1.603) = 100.0.03365 / = 23,63
1.80 0.01091 1.85 0.00889 1.90 0.00721 1.95 0.00582 2.00 0.00468 2.05 0.00374	of a clay liner (thickness ~ 90 cms) at the end of 5 years is approximately:  (a. 20-25 ppb b. 200-250 ppb c. 2-5 ppb d. ~1000ppb  (a. 1000 ppb c. 2-5 ppb d. ~1000ppb  (b. 200-250 ppb c. 2-5 ppb d. ~1000ppb  (c) = 1000  (c
2.10 0.00298	6. While discussing with your colleagues the suitability of this liner you suggest mixing "zeolitee" with your
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2.10 0.00298 2.15 0.00236 2.20 0.00186 2.25 0.00146 2.36 0.00189 2.40 0.00069 2.45 0.00031 2.50 0.00031 2.50 0.00018 2.70 0.00013 2.75 0.00010 2.80 0.00008 2.85 0.00008 2.85 0.00006 2.90 0.00004 2.95 0.00003	6. While discussing with your colleagues the suitability of this liner you suggest mixing "zeolites" with your clay material before compacting the liner. You know that the contaminant will be adsorbed into the zeolite particles as it travels through the liner and preliminary tests suggest a retardation factor of 2 (R=2). All the other parameters remain the same. It is presumed that the addition of zeolites will not change the hydraulic conductivity, the constrained molecular diffusion or the dispersivity. The longitudinal dispersivity for this liner is 1 cm. With this new design the concentration of the above chemical (i.e., C= 1000 ppb) at the bottom of a clay liner (thickness ~ 90 cms) at the end of 5 years, based on advection, dispersion and diffusion is approximately:  a. same as before  b. 200-250 ppb  c. 20-50 ppb  d. 1-2 ppb  C. 3333  Quantam  Oct (10-2)  Mixing Quan  Oct (10-2)  Mixing Quan  Oct (10-2)
2.10 0.00298 2.15 0.00236 2.20 0.00186 2.25 0.00146 2.36 0.00114 2.35 0.00089 2.45 0.00053 2.50 0.00041 2.55 0.00031 2.60 0.00024 2.65 0.00018 2.70 0.00013 2.75 0.00010 2.80 0.00008 2.85 0.00006 2.90 0.00004 2.95 0.00003	6. While discussing with your colleagues the suitability of this liner you suggest mixing "zeolites" with your clay material before compacting the liner. You know that the contaminant will be adsorbed into the zeolite particles as it travels through the liner and preliminary tests suggest a retardation factor of 2 (R=2). All the other parameters remain the same. It is presumed that the addition of zeolites will not change the hydraulic conductivity, the constrained molecular diffusion or the dispersivity. The longitudinal dispersivity for this liner is 1 cm. With this new design the concentration of the above chemical (i.e., C <sub>10</sub> = 1000 ppb) at the bottom of a clay liner (thickness - 90 cms) at the end of 5 years, based on advection, dispersion and diffusion is approximately:  a. same as before  b. 200-250 ppb  c. 20-50 ppb  d. 1-2 ppb  C. 20-50 ppb  d. 1-2 ppb  C. 20-333 (30-m/a)  C. 20-50 ppb  d. 1-2 ppb  C. 20-50 ppb  d. 20-60 ppb  d. 20
2.10 0.00298 2.15 0.00236 2.20 0.00186 2.25 0.00146 2.36 0.00114 2.35 0.00089 2.45 0.00053 2.50 0.00041 2.55 0.00031 2.60 0.00024 2.65 0.00018 2.70 0.00013 2.75 0.00010 2.80 0.00008 2.85 0.00006 2.90 0.00004 2.95 0.00003	6. While discussing with your colleagues the suitability of this liner you suggest mixing "zeolites" with your clay material before compacting the liner. You know that the contaminant will be adsorbed into the zeolite particles as it travels through the liner and preliminary tests suggest a retardation factor of 2 (R=2). All the other parameters remain the same. It is presumed that the addition of zeolites will not change the hydraulic conductivity, the constrained molecular diffusion or the dispersivity. The longitudinal dispersivity for this liner is 1 cm. With this new design the concentration of the above chemical (i.e., C <sub>10</sub> = 1000 ppb) at the bottom of a clay liner (thickness - 90 cms) at the end of 5 years, based on advection, dispersion and diffusion is approximately:  a. same as before  b. 200-250 ppb  c. 20-50 ppb  d. 1-2 ppb  C. 20-50 ppb  d. 1-2 ppb  C. 20-333 (30-m/a)  C. 20-50 ppb  d. 1-2 ppb  C. 20-50 ppb  d. 20-60 ppb  d. 20
2.10 0.00298 2.15 0.00236 2.20 0.00186 2.25 0.00146 2.36 0.00114 2.35 0.00089 2.45 0.00053 2.50 0.00041 2.55 0.00031 2.60 0.00024 2.65 0.00018 2.70 0.00013 2.75 0.00010 2.80 0.00008 2.85 0.00006 2.90 0.00004 2.95 0.00003	6. While discussing with your colleagues the suitability of this liner you suggest mixing "zeolites" with your clay material before compacting the liner. You know that the contaminant will be adsorbed into the zeolite particles as it travels through the liner and preliminary tests suggest a retardation factor of 2 (R=2). All the other parameters remain the same. It is presumed that the addition of zeolites will not change the hydraulic conductivity, the constrained molecular diffusion or the dispersivity. The longitudinal dispersivity for this liner is 1 cm. With this new design the concentration of the above chemical (i.e., C <sub>0</sub> = 1000 ppb) at the bottom of a clay liner (thickness ~ 90 cms) at the end of 5 years, based on advection, dispersion and diffusion is approximately:  a. same as before  b. 200-250 ppb  c. 20-50 ppb  d. 1-2 ppb  C. 3333  G. 3333  G. 302 m 190 m  G. 3333  G. 3333  G. 302 m 190 m  G. 3333